PHYSICAL SCIENCE (PHSC-112) GRADE 11 - WEEK 1-10 AMALEAKS.BLOGSPOT.COM

 What do you call the solid formed when a liquid solution is allowed to stand still for a long amount of time, ans. Precipitate

2. What is the proper term for catalysts involved in biological processes? ans. Enzymes

 What is the term for the energy responsible for the variations in the reaction rates of chemical processes? answer:activation energy

4. What property of a material describes the compactness of the arrangement of its molecules? ans.Density

 Identify the kind of change wherein the product is the same as the reactants. ANWERophysical change

6. In which type of covalent bonding are the shared electrons shared congruently throughout the molecule? answer:nonpolar

What do you call the ability of a substance to chemically combine with other compounds? answer:Reactivity

8.Fill in the blank: The __

property of a substance can be observed without changing the substance, answer:physical

What type of covalent compound is there a slight difference in the electric charge between the opposite sides of the molecule? answer:polar covalent compound

50. Identification: This factor in a reaction rate is a consequence of the kinetic energy of the reactant molecules? wrong answer ako dito.

11. What kind of chemical bond involves the sharing of electrons between two atoms? answer:covalent bond 12.What is the component of a chemical reaction that is present at its start? answer:reactant 13.Identification: Two different atoms may or may not have this type of covalent bonding, answer:polar covalent bonding.

14 _____covalent bond does not have an equal sharing of electrons, answer.Polar

15. What kind of change results to the production of a material that is entirely different from the reactants? answer:chemical change

16. Identification: This is another term used to refer to covalent compounds, answer:molecular compound 17. Identification: This refers to the speed of a chemical process, answer:reaction rate 18. Which atom is more positive than the other in a molecule of dihydrogen oxide? answer: hydrogen atom 19. Identification: These compounds are highly soluble in water, answersionic compounds.

20.Identification: These properties can be detected using your senses, answer:Physical properties

21.What type of compound is formed by the sharing of electrons? answer:Covalent compound

22.covalent bonding exists between two identical atoms, answer; nonpolar

23 Identification: This is the force that exists between the molecules of a compound, answer: Intermolecular force

24. What is the term for the property that makes it capable of being hammered into thin sheets? answer:malleability

25. In a water molecule, which atom is more frequented by the shared electrons? answer:oxygen 26.Identify the substance that hastens chemical reactions? answer:catalyst 27. In a

type of covalent bonding, there is equal sharing of electrons between the atoms, answer; nonpolar 28. What type of covalent bonding is characterized by the incongruent sharing of electrons between the

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ML Yell

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Resources in Education ,1997 International Conference on Science Education 2012 Proceedings Baohui Zhang, Gavin W. Fulmer, Xiufeng Liu, Weiping Hu, Shujin Peng, Bing Wei, 2014-05-06 This book contains papers presented at the International Conference on Science Education 2012 ICSE 2012 held in Nanjing University Nanjing China It features the work of science education researchers from around the world addressing a common theme Science Education Policies and Social Responsibilities The book covers a range of topics including international science education standards public science education and science teacher education It also examines how STEM education has dominated some countries science education policy ways brain research might provide new approaches for assessment how some countries are developing their new national science education standards with research based evidence and ways science teacher educators can learn from each other Science education research is vital in the development of national science education policies including science education standards teacher professional development and public understanding of science Featuring the work of an international group of science education researchers this book offers many insightful ideas experiences and strategies that will help readers better understand and address challenges in the field **Chemical News and Journal of Physical Reasoning in Physics** L. Viennot, 2007-05-08 For a meaningful understanding of physics it is necessary **Science** ,1890 to realise that this corpus of knowledge operates in a register different from natural thought This book aims at situating the main trends of common reasoning in physics with respect to some essential aspects of accepted theory. It analyses a great many research results based on studies of pupils and students at various academic levels involving a range of physical situations It shows the impressive generality of the trends of common thought as well as their resistance to teaching The book s main focus is to underline to what extent natural thought is organised As a result of this mapping out of trends of reasoning some suggestions for teaching are presented these have already influenced recent curricula in France This book is intended for teachers and teacher trainers principally but students can also benefit from it to improve their understanding of Title Index, physics and of their own ways of reasoning The Chemical News and Journal of Physical Science, 1889 **ERIC ED Accessions File** ERIC Processing and Reference Facility, 1966 Research in Education ,1973 Secondary **School Science Teaching Practices** Horatio Seymour Fowler,1964 Catalog of Copyright Entries. Third Series Library **Physics for You** Keith Johnson, 2001 The most popular series for GCSE has been of Congress. Copyright Office, 1977 updated to offer comprehensive coverage of the revised GCSE specifications Physics for You has been updated in line with the revised National Curriculum requirements

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