

Chapter 2 Atoms and the Periodic Table

Solutions to In-Chapter Problems

2.1 Each element is identified by a one- or two-letter symbol. Use the periodic table to find the symbol for each element.

- a. Ca b. Rn c. N d. Au

2.2 Use the periodic table to find the symbol for each element.

- a. Cu and Zn b. Cu and Sn c. Sn, Sb, and Pb

2.3 Use the periodic table to find the element corresponding to each symbol.

- a. neon b. sulfur c. iodine d. silicon e. boron f. mercury

2.4 Metals are shiny materials that are good conductors of heat and electricity. Nonmetals do not have a shiny appearance, and they are generally poor conductors of heat and electricity. Metalloids have properties intermediate between metals and nonmetals.

- a, d, f, h: metals b, c, g: nonmetals e: metalloid

2.5 Use Figure 2.1 and the definitions in Answer 2.4 to determine if the micronutrients are metals, nonmetals, or metalloids.

- As, B, Si: metalloids Cr, Co, Cu, Fe, Mn, Mo, Ni, Zn: metals F, I, Se: nonmetals

2.6 Use Figure 2.3 to determine which elements are represented in the molecular art.

- a. 4 hydrogens, 1 carbon b. 3 hydrogens, 1 nitrogen c. 6 hydrogens, 2 carbons, 1 oxygen

2.7 The subscript tells how many atoms of a given element are in each chemical formula.

- | | |
|--|--|
| a. NaCN (sodium cyanide) = 1 sodium, 1 carbon, 1 nitrogen | d. SnF ₂ (stannous fluoride) = 1 tin, 2 fluorines |
| b. H ₂ S (hydrogen sulfide) = 2 hydrogens, 1 sulfur | e. CO (carbon monoxide) = 1 carbon, 1 oxygen |
| c. C ₂ H ₆ (ethane) = 2 carbons, 6 hydrogens | f. C ₃ H ₈ O ₃ (glycerol) = 3 carbons, 8 hydrogens, 3 oxygens |

2.8 Use Figure 2.3 to determine which elements are represented in the molecular art.

Halothane contains 2 carbons, 1 hydrogen, 3 fluorines, 1 bromine, and 1 chlorine atom.

2.9

- a. In a neutral atom, the number of protons and electrons is equal; 9 protons = 9 electrons.
b. The atomic number = the number of protons = 9.
c. This element is fluorine.

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