

ACIDS AND BASES QUIZ #2

Multiple Choice: Select the most correct answer. (15 E)

1. Which one is correctly matched?

- A) Acids - pH range above 7
- B) Acids - pH range below 7
- C) Acids - pH range 7 (neutral)
- D) Acids - pH range 8-9

2. Which of the following is NaOH a strong base?

- A) NaOH_{aq} B) KOH_{aq} C) LiOH_{aq}
- D) CaOH_{aq} E) $\text{Mg(OH)}_{\text{aq}}$

3. Which of the following equations represents an acid-base neutralization reaction?

- A) $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$
- B) $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$
- C) $\text{Ba(OH)}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$
- D) more than one correct response
- E) none of the above

4. In which of the following pairs of acids are both members of the pair strong acids?

- A) HNO_3 and HNO_2
- B) H_2SO_4 and H_2SO_3
- C) HCl and HClO_4
- D) more than one correct response
- E) none of the above

5. Which of the following does not apply to the pH scale?

- A) The range of numbers is 0 to 14.
- B) Acids on the scale range from 0 to 14.
- C) A pH of 9 indicates a base
- D) Stomach acid would have a pH less than 7.

6. An acid solution has a pH of 4.0. If water is added to this solution to increase its volume, which of the following will occur?

- A) The pH will decrease
- B) The pH will increase
- C) $[\text{H}^+]_{\text{aq}}$ will increase
- D) $[\text{H}^+]_{\text{aq}}$ will decrease
- E) b and d only

7. The acidity of carbonated water is due to the _____.

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PT Brinkman



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