

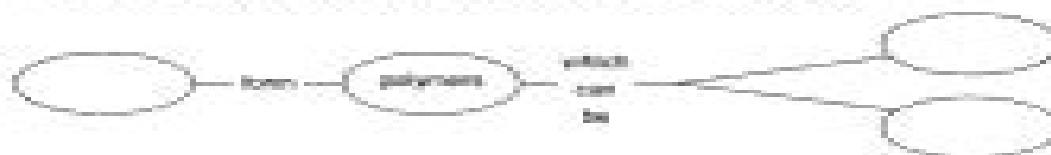
Chapter 9 Carbon Chemistry

Section 9.3 Polymers**(pages 275–280)**

This section explains how polymers form. It also discusses examples of synthetic and natural polymers.

Reading Strategy *page 275*

Identifying Main Ideas As you read, complete the concept map to summarize two main ideas about polymers. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.



1. Describe a polymer. _____
2. The smaller molecules that join together to form a polymer are called _____.
3. Is the following sentence true or false? More than one type of monomer can be present in some polymers. _____
4. Name the two general classifications of polymers.
 - a. _____
 - b. _____

Synthetic Polymers *(page 276)*

5. Name three polymers that can be synthesized.
 - a. _____
 - b. _____
 - c. _____
6. Is the following sentence true or false? The more carbon atoms there are in a polyethylene chain, the harder the polymer is. _____

Natural Polymers *(pages 276–280)*

7. Name four types of polymers that are produced in plant and animal cells.
 - a. _____
 - b. _____
 - c. _____
 - d. _____
8. Circle the letter of the molecular formula of a simple sugar.
 - a. CH_4O
 - b. $\text{C}_6\text{H}_{12}\text{O}_6$
 - c. $\text{C}_{12}\text{H}_{22}\text{O}_{11}$
 - d. $\text{C}_{60}\text{H}_{12}\text{O}_{12}$

Section 93 Polymers Answers

Marco Cascella



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