



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Padhraic Smyth



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Table of Contents Physical Science Grade 11 2014 First Term Paper Wiki

1. Understanding the eBook Physical Science Grade 11 2014 First Term Paper Wiki
 - The Rise of Digital Reading Physical Science Grade 11 2014 First Term Paper Wiki
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 11 2014 First Term Paper Wiki
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 11 2014 First Term Paper Wiki
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 11 2014 First Term Paper Wiki
 - Personalized Recommendations
 - Physical Science Grade 11 2014 First Term Paper Wiki User Reviews and Ratings
 - Physical Science Grade 11 2014 First Term Paper Wiki and Bestseller Lists

5. Accessing Physical Science Grade 11 2014 First Term Paper Wiki Free and Paid eBooks
 - Physical Science Grade 11 2014 First Term Paper Wiki Public Domain eBooks
 - Physical Science Grade 11 2014 First Term Paper Wiki eBook Subscription Services
 - Physical Science Grade 11 2014 First Term Paper Wiki Budget-Friendly Options
6. Navigating Physical Science Grade 11 2014 First Term Paper Wiki eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade 11 2014 First Term Paper Wiki Compatibility with Devices
 - Physical Science Grade 11 2014 First Term Paper Wiki Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade 11 2014 First Term Paper Wiki
 - Highlighting and Note-Taking Physical Science Grade 11 2014 First Term Paper Wiki
 - Interactive Elements Physical Science Grade 11 2014 First Term Paper Wiki
8. Staying Engaged with Physical Science Grade 11 2014 First Term Paper Wiki
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 11 2014 First Term Paper Wiki
9. Balancing eBooks and Physical Books Physical Science Grade 11 2014 First Term Paper Wiki
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 11 2014 First Term Paper Wiki
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade 11 2014 First Term Paper Wiki
 - Setting Reading Goals Physical Science Grade 11 2014 First Term Paper Wiki
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade 11 2014 First Term Paper Wiki
 - Fact-Checking eBook Content of Physical Science Grade 11 2014 First Term Paper Wiki
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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