

Physical Science: Element Worksheet

Name _____ Date _____

Multiple Choice

- _____ 1. Mendeleev arranged the elements in his periodic table in order of...
a. atomic number
b. number of electrons
c. mass
d. number of neutrons
- _____ 2. Mendeleev's decision to leave gaps in his periodic table was supported by the discovery of...
a. electrons
b. protons
c. argon
d. gallium
- _____ 3. In Mendeleev's periodic table, elements with similar properties were grouped...
a. in the same row
b. in pairs of two
c. in the same column
d. along diagonal lines that run from left to bottom right
- _____ 4. Which of the following is not a characteristic of metals?
a. ductile
b. typically solid at room temp.
c. good conductors of electricity
d. brittle
- _____ 5. The elements from group 2 on the periodic table are called...
a. Halogens
b. Alkaline Earth Metals
c. Alkali Metals
d. Noble Gases
- _____ 6. Why do elements in a group have similar chemical properties?
a. They have the same number of valence electrons.
b. They have the same ratio of protons to neutrons.
c. Their atomic masses are multiples of the least massive element of the group.
d. They have the same number of total electrons.
- _____ 7. How many periods does the modern periodic table have?
a. 5
b. 7
c. 10
d. 18
- _____ 8. Which elements are all gases at room temp. (25° C)?
a. Group 1
b. Group 7
c. Group 8
d. Group 18
- _____ 9. The most reactive metals are the...
a. Transition metals
b. Alkali metals
c. metalloids
d. Noble gases
- _____ 10. An element that is shiny and conducts electricity is likely to be a...
a. Gas
b. metalloid
c. liquid
d. metal
- _____ 11. Moving up and down (vertically) on the periodic table would represent which of the following best?
a. Group
b. Family
c. Period
d. A and B

Physical Science Element Answer Key

H Kauffman



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Table of Contents Physical Science Element Answer Key

1. Understanding the eBook Physical Science Element Answer Key
 - The Rise of Digital Reading Physical Science Element Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Element Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Physical Science Element Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Element Answer Key
 - Personalized Recommendations

- Physical Science Element Answer Key User Reviews and Ratings
- Physical Science Element Answer Key and Bestseller Lists
- 5. Accessing Physical Science Element Answer Key Free and Paid eBooks
 - Physical Science Element Answer Key Public Domain eBooks
 - Physical Science Element Answer Key eBook Subscription Services
 - Physical Science Element Answer Key Budget-Friendly Options
- 6. Navigating Physical Science Element Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Element Answer Key Compatibility with Devices
 - Physical Science Element Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Element Answer Key
 - Highlighting and Note-Taking Physical Science Element Answer Key
 - Interactive Elements Physical Science Element Answer Key
- 8. Staying Engaged with Physical Science Element Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Element Answer Key
- 9. Balancing eBooks and Physical Books Physical Science Element Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Element Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Element Answer Key
 - Setting Reading Goals Physical Science Element Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Element Answer Key
 - Fact-Checking eBook Content of Physical Science Element Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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