



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

Sekhukhune Life Science Question Paper Term 1

Memorandum

Michael Brown



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Enjoying the Tune of Term: An Emotional Symphony within **Sekhukhune Life Science Question Paper Term 1 Memorandum**

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Table of Contents Sekhukhune Life Science Question Paper Term 1 Memorandum

1. Understanding the eBook Sekhukhune Life Science Question Paper Term 1 Memorandum
 - The Rise of Digital Reading Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Advantages of eBooks Over Traditional Books
2. Identifying Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sekhukhune Life Science Question Paper Term 1 Memorandum
 - User-Friendly Interface

4. Exploring eBook Recommendations from Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Personalized Recommendations
 - Sekhukhune Life Science Question Paper Term 1 Memorandum User Reviews and Ratings
 - Sekhukhune Life Science Question Paper Term 1 Memorandum and Bestseller Lists
5. Accessing Sekhukhune Life Science Question Paper Term 1 Memorandum Free and Paid eBooks
 - Sekhukhune Life Science Question Paper Term 1 Memorandum Public Domain eBooks
 - Sekhukhune Life Science Question Paper Term 1 Memorandum eBook Subscription Services
 - Sekhukhune Life Science Question Paper Term 1 Memorandum Budget-Friendly Options
6. Navigating Sekhukhune Life Science Question Paper Term 1 Memorandum eBook Formats
 - ePub, PDF, MOBI, and More
 - Sekhukhune Life Science Question Paper Term 1 Memorandum Compatibility with Devices
 - Sekhukhune Life Science Question Paper Term 1 Memorandum Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Highlighting and Note-Taking Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Interactive Elements Sekhukhune Life Science Question Paper Term 1 Memorandum
8. Staying Engaged with Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sekhukhune Life Science Question Paper Term 1 Memorandum
9. Balancing eBooks and Physical Books Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sekhukhune Life Science Question Paper Term 1 Memorandum
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Setting Reading Goals Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Fact-Checking eBook Content of Sekhukhune Life Science Question Paper Term 1 Memorandum
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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