

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the chemical equation
- 2. Explain how to interpret the equation
- 3. List the steps for writing a chemical equation

Vocabulary

- 1. chemical equation
- 2. chemical equation
- 3. reactant
- 4. coefficient
- 5. balanced equation

Part A Completion

Write the coefficient for each in the chemical equation, and balance the chemical equation. Write the name of each substance in the equation.

_____ a chemical equation is always represented by a balanced _____
 The chemical equation is a chemical change called _____ that uses
 chemical substances (reactants) to make _____ products. The
 new substances are _____ a chemical equation is written with an
 arrow pointing right, and plus _____ is used to separate the
 substances. The left number of atoms of an _____ atom is found in
 the equation. The equation must always balance by changing the _____ of
 the chemical equation.

Chemical equations are written with the chemical formula as a
 symbol. The equation is _____ that a plus _____ is used to separate
 the _____ substances. The equation is balanced _____ of _____ of
 each element. The equation is balanced by changing the _____ of the
 equation.

- 1. equation
- 2. reaction
- 3. products
- 4. plus
- 5. coefficients
- 6. atoms
- 7. balanced
- 8. _____
- 9. _____
- 10. _____
- 11. _____
- 12. _____
- 13. _____
- 14. _____
- 15. _____

Section Review 11 1 Describing Chemical Reactions

Jiyuan Zhang



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