



Science Curriculum

Chemical Changes

Read the passage and answer the following questions.



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide and the products are water and oxygen. The chemical formula for hydrogen peroxide is H_2O_2 . It is a colorless liquid with a strong odor. It is used in many household products, such as bleach and hair dye.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

Hydrogen peroxide is a colorless liquid with a strong odor. It is used in many household products, such as bleach and hair dye. It is a chemical compound with the formula H_2O_2 . It is a liquid at room temperature and pressure. It is a colorless liquid with a strong odor. It is used in many household products, such as bleach and hair dye. It is a chemical compound with the formula H_2O_2 . It is a liquid at room temperature and pressure. It is a colorless liquid with a strong odor. It is used in many household products, such as bleach and hair dye.

4. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

It is equal.



5. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

6. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

7. What is the chemical formula for hydrogen peroxide? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

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10. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

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WJ Hussar



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