

Name _____

Development of Modern Atomic Theory – Video worksheet

1. The development of atomic theory started over 2000 years ago with a man named _____.
2. He believed that all matter is made up of tiny particles called _____.
3. Who was the next person to advance atomic theory in 1808? _____
4. He thought that atoms of a given element are identical in _____, _____, and other properties. Today we know this to NOT be true because of _____.
5. 90 years later, a man named J.J. Thompson discovered the _____ in 1897.
6. His model of the atom is referred to as the _____ model.
7. A Crooke's tube is essentially the same as a _____ tube.
8. Ernest Rutherford won the nobel prize in 1908 for his discovery of the _____.
9. His famous experiment is known as the _____ experiment.
10. Rutherford's model of the atom is known as the _____ model of the atom.
11. What are the tiny particles in the nucleus of the atom? _____
12. Who was the Danish physicist who won the nobel prize in 1922? _____
13. In his model, electrons move around the nucleus of the atom in a _____ motion.
14. In the Bohr model, a _____ is emitted with energy $E = hf$ when an electron jumps down from one orbit to another.
15. Louis de Broglie was a French physicist who made groundbreaking contributions to _____ theory.
16. He postulated that all matter exhibits _____ properties.
17. Our current understanding of the atom is based on the _____ Wave Mechanical Model of the Atom.
18. The Heisenberg Uncertainty Principle states that no experiment can measure both the _____ and _____ of a quantum particle simultaneously.
19. The Schrodinger wave equation shows that atomic particles like _____ exhibit wave-like behaviors.
20. The wave mechanical model of the atom shows the _____ of finding an electron in an atom.

Section Review Development Of The Atomic Theory Answers

Neil D. Jespersen, Alison Hyslop



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Table of Contents Section Review Development Of The Atomic Theory Answers

1. Understanding the eBook Section Review Development Of The Atomic Theory Answers
 - The Rise of Digital Reading Section Review Development Of The Atomic Theory Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section Review Development Of The Atomic Theory Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section Review Development Of The Atomic Theory Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section Review Development Of The Atomic Theory Answers
 - Personalized Recommendations
 - Section Review Development Of The Atomic Theory Answers User Reviews and Ratings
 - Section Review Development Of The Atomic Theory Answers and Bestseller Lists
5. Accessing Section Review Development Of The Atomic Theory Answers Free and Paid eBooks
 - Section Review Development Of The Atomic Theory Answers Public Domain eBooks
 - Section Review Development Of The Atomic Theory Answers eBook Subscription Services
 - Section Review Development Of The Atomic Theory Answers Budget-Friendly Options
6. Navigating Section Review Development Of The Atomic Theory Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section Review Development Of The Atomic Theory Answers Compatibility with Devices
 - Section Review Development Of The Atomic Theory Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section Review Development Of The Atomic Theory Answers
 - Highlighting and Note-Taking Section Review Development Of The Atomic Theory Answers
 - Interactive Elements Section Review Development Of The Atomic Theory Answers
8. Staying Engaged with Section Review Development Of The Atomic Theory Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Section Review Development Of The Atomic Theory Answers
- 9. Balancing eBooks and Physical Books Section Review Development Of The Atomic Theory Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section Review Development Of The Atomic Theory Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section Review Development Of The Atomic Theory Answers
 - Setting Reading Goals Section Review Development Of The Atomic Theory Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section Review Development Of The Atomic Theory Answers
 - Fact-Checking eBook Content of Section Review Development Of The Atomic Theory Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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