

Review for Ch. 8 Test

8.1 Describing Chemical Reactions

- know how to write word equations (i.e. you have be able to name compounds, including acids, ionic, and covalent compounds from Ch. 7)
- know how to write a formula equation
- What 2 things do coefficients represent in an equation? **Relative number of moles of a substance ; relative number of molecules needed for a reaction to occur**
- know common symbols used in a reaction
- know 4 evidences of a chemical reaction:
 1. **energy in the form of heat or light**
 2. **production of a gas**
 3. **change in color**
 4. **formation of a precipitate**
- know the Law of conservation of matter: **matter can neither be created nor destroyed in a chemical reaction (which means, mass of reactants must equal the mass of products)**
- know what a chemical equation represents: **relative amounts of reactants and products**
- know how to balance chemical equations (combustion reactions are particularly tricky, practice!)

1. Complete the equation and write a word equation for the following reaction: $\text{ZnO} + \text{FeSO}_4 \rightleftharpoons$



Zinc oxide reacts with iron (II) sulfate to yield zinc sulfate and iron (II) oxide

2. Write a formula equation for the reaction of a solution of AgCl with potassium metal:



3. What elements are always diatomic gases at room temperature?

All of the halogens (Group 17) , H_2 , N_2 , O_2

4. Can you balance these reactions:



Section Review 1describing Chemical Reactions Answer Key

Camilla Rothe



Section Review 1describing Chemical Reactions Answer Key:

Holt Chemistry Ralph Thomas Myers,2004 *Modern Chemistry* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Applied Mechanics Reviews** ,1973 **Radioactive Waste Management** ,1990-07 Chemistry 'O' Level Rex M. Heyworth,2007 Transport and Fate of Chemicals in the Environment John S. Gulliver,2012-12-12 What happens when a chemical is released into the environment It diffuses disperses adsorbs reacts and or changes state To predict and analyze this process the mathematics of diffusion is applied to lakes rivers groundwater the atmosphere the oceans and transport between these media A sustainable world requires a deep understanding of the transport of chemicals through the environment and how to address and harness this process This volume presents a succinct and in depth introduction to this critical topic Featuring authoritative peer reviewed articles from the Encyclopedia of Sustainability Science and Technology *Transport and Fate of Chemicals in the Environment* represents an essential one stop reference for an audience of researchers undergraduate and graduate students and industry professionals *Columbia Review High-yield General Chemistry* Stephen D. Bresnick,1996 A study guide in question and answer format for basic chemistry **Essentials for Health and Wellness** Gordon Edlin,Eric Golanty,Kelli McCormack Brown,2000 Health Wellness **Cehmistry Textbook for College and University USA** Ibrahim Sikder,2023-06-04 Cehmistry Textbook USA **The Sciences** James Trefil,Robert M. Hazen,2016-09-13 This text is an unbound three hole punched version The Sciences An Integrated Approach Binder Ready Version 8th Edition by James Trefil and Robert Hazen uses an approach that recognizes that science forms a seamless web of knowledge about the universe This text fully integrates physics chemistry astronomy earth sciences and biology and emphasizes general principles and their application to real world situations The goal of the text is to help students achieve scientific literacy Applauded by students and instructors for its easy to read style and detail appropriate for non science majors the eighth edition has been updated to bring the most up to date coverage to the students in all areas of science Supercritical Carbon Dioxide Maartje F. Kemmere,Thierry Meyer,2006-05-12 Recently supercritical fluids have emerged as more sustainable alternatives for the organic solvents often used in polymer processes This is the first book emphasizing the potential of supercritical carbon dioxide for polymer processes from an engineering point of view It develops a state of the art overview on polymer fundamentals polymerization reactions and polymer processing in supercritical carbon dioxide The book covers topics in a multidisciplinary approach starting from polymer chemistry and thermodynamics going through monitoring polymerization processes and ending with polymer shaping and post processing The authors are internationally recognized experts from different fields in polymer reaction engineering in supercritical fluids The book was initiated by the Working Party on Polymer Reaction Engineering of the European Federation of Chemical Engineering and further renowned international experts Prentice Hall Science Explorer Michael J. Padilla,2002 Chemical Reactivity in Confined Systems Pratim Kumar Chattaraj,Debdutta Chakraborty,2021-08-13 An insightful analysis of confined chemical

systems for theoretical and experimental scientists

Chemical Reactivity in Confined Systems Theory and Applications presents a theoretical basis for the molecular phenomena observed in confined spaces The book highlights state of the art theoretical and computational approaches with a focus on obtaining physically relevant clarification of the subject to enable the reader to build an appreciation of underlying chemical principles The book includes real world examples of confined systems that highlight how the reactivity of atoms and molecules change upon encapsulation Chapters include discussions on recent developments related to several host guest systems including cucurbit n uril ExBox 4 clathrate hydrates octa acid cavitand metal organic frameworks MOFs covalent organic frameworks COFs zeolites fullerenes and carbon nanotubes Readers will learn how to carry out new calculations to understand the physicochemical behavior of confined quantum systems Topics covered include A thorough introduction to global reactivity descriptors including electronegativity hardness and electrophilicity An exploration of the Fukui function as well as dual descriptors higher order derivatives and reactivity through information theory A practical discussion of spin dependent reactivity and temperature dependent reactivity Concise treatments of population analysis reaction force electron localization functions and the solvent effect on reactivity Perfect for academic researchers and graduate students in theoretical and computational chemistry and confined chemical systems **Chemical Reactivity in Confined Systems Theory and Applications** will also earn a place in the libraries of professionals working in the areas of catalysis supramolecular chemistry and porous materials

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