9-2 Review and Reinforcement

Chemical Equations

Refer to the unbalanced chemical equation below to answer each of the following questions. Write your answers in the space provided.

Z AKNO, 1, +3 FeCl, -3 Fe(NO,) +ZAICI,

1. Write the names of the reactants in this equation.

Alzenia som Witterte + Tran (11) chloride

2. Write the names of the products in this equation.

From (a) Niverty + Aluminum chloride

- What would the coefficient of Al(NO₂)₃ be in the balanced equation? TWO
- What would the coefficient of Fe(NO₃), be in the balanced equation? The ext.
- 6. How many oxygen atoms would there be on each side of the balanced equation? 18.

Answer each of the following questions in the space provided.

7. What should you do if an equation you have written cannot be balanced?

Correctly, tric again, and for help

8. How is balancing a chemical reaction related to the law of conservation of matter?

mass of reachists must = muss of products

9. Describe the correct way to balance an equation.

and only coefficients Never change Comulas

10. How does a chemical equation relate the macroscopic, observable reaction to the interactions occurring at the microscopic level?

Some

Section 3 Reinforcement Transition Elements Paper Answers

SJ Ball

Section 3 Reinforcement 7	Transition	${\bf Elements}$	Paper	Answers:
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