

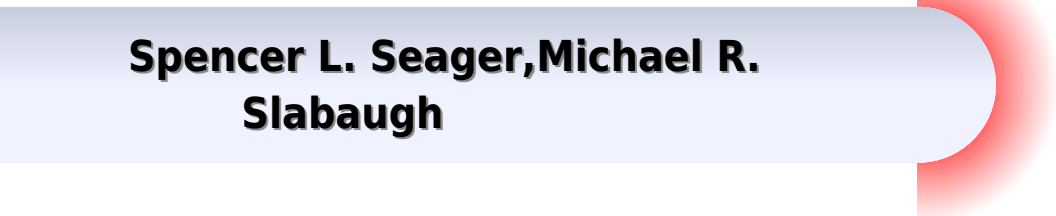
Study Guide Section 11.3 Limiting Reactants**Key Concepts**

- The limiting reactant is the reactant that is completely consumed during a chemical reaction. Reactants that remain after the reaction stops are called excess reactants.
- To determine the limiting reactant, the actual mole ratio of the available reactants must be compared with the ratio of the reactants obtained from the coefficients in the balanced chemical equation.
- Stoichiometric calculations must be based on the limiting reactant.



Section 3 Limiting Reactants Key

**Spencer L. Seager, Michael R.
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