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CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

$2H_0 + O_0 \rightarrow 2H_0O$	Caco, Cao + Co,	
3K + 3H ₂ O → H ₄ + 3KOH	CHOO, • 2HCI CHOL • HLCO.	
290010 ₉ 28001 + 30 ₉	Ca + 2HCl CaCl ₂ + H ₂	
280 _e + 0 _e 280 _e	PERNOJ, - SHCI - PECL - SHEED,	
$Ov + 2AgNO_y \rightarrow 2Ag + OutyOy$	2Na • Cl _e 2NeCli	
2H ₂ O 2H ₂ - O ₂	CH ₆ + 20 ₅ → CO ₃ + 24 ₅ O	
$PbONO_{\mathcal{O}_{S}} = 290 \rightarrow PbI_{b} = 29000_{b}$	H,CO, CO, - H,O	
2Fe + O₂ → 2FeO	Zn + 2HCt H _b + ZnCt _b	
HCI + NaOH → NaOI + HOH	25% + 254,0 25%0H + H ₂	

Physical Science If8767 Classifying Chemical Reactions

LL Leslie

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