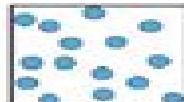


The Three States of Matter

Grade 5 - Properties and Changes in Matter

Handout - Answers

1. Draw the particles of the three different states of matter (solid, liquid, and gas).

Solid	Liquid	Gas
		

2. Provide an example of each state of matter discussed:

Solid: **Any solid (e.g. Chair)** Liquid: **Any liquid (e.g. juice)** Gas: **Any gas (e.g. Oxygen)**

3. At what temperature does water turn into a solid? **0°C**
4. At what temperature does water turn into a gas? **100°C**
5. Based on the experiments conducted do you think all liquids turn to solid at the same temperature? Do you think all liquids turn to gas at the same temperature? Explain.
(Based on any experiment) **Different liquids turn to gas at different temperatures. An example would be that the sugar crystals turned back into a solid even though it was not 0°C or below.**
6. Once something is a solid, does it stay that way forever? Explain.
***Hint: Think about the ice cream experiment and use it as an example.**
No, it can change back to being a liquid (and possibly a gas). For example, we made ice cream (a solid) out of liquid cream but when I left it out too long some of it started to melt and turn back into a liquid
7. Challenge Question: If sugar is a solid then why does it take the shape of the container it is put in? What about with salt or sand?
Sugar (as well as salt and sand) is still a solid. Each crystal has a fixed shape and volume, does not flow under shear stress, does not take the shape of its container, and is incompressible. Many small solids (grains) moving together can sometimes act like a liquid (called granular flow).

Section States Of Matter Answers

R Barnett



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Table of Contents Section States Of Matter Answers

1. Understanding the eBook Section States Of Matter Answers
 - The Rise of Digital Reading Section States Of Matter Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section States Of Matter Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section States Of Matter Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section States Of Matter Answers
 - Personalized Recommendations

- Section States Of Matter Answers User Reviews and Ratings
- Section States Of Matter Answers and Bestseller Lists
- 5. Accessing Section States Of Matter Answers Free and Paid eBooks
 - Section States Of Matter Answers Public Domain eBooks
 - Section States Of Matter Answers eBook Subscription Services
 - Section States Of Matter Answers Budget-Friendly Options
- 6. Navigating Section States Of Matter Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section States Of Matter Answers Compatibility with Devices
 - Section States Of Matter Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section States Of Matter Answers
 - Highlighting and Note-Taking Section States Of Matter Answers
 - Interactive Elements Section States Of Matter Answers
- 8. Staying Engaged with Section States Of Matter Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section States Of Matter Answers
- 9. Balancing eBooks and Physical Books Section States Of Matter Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section States Of Matter Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section States Of Matter Answers
 - Setting Reading Goals Section States Of Matter Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section States Of Matter Answers
 - Fact-Checking eBook Content of Section States Of Matter Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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