

2.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using energy flow data.
3. Apply the law of conservation of energy to chemical reactions.

Vocabulary

- | | | |
|------------------|---------------|----------------------------|
| 1. endothermic | 2. exothermic | 3. enthalpy |
| 4. heat capacity | 5. joule | 6. thermochemical equation |

Part A Completion

Use the vocabulary words to identify a vocabulary word that matches each of the following phrases. Write the word in the space provided.

- | | |
|---|-----------|
| 1. A process that absorbs energy is called a(n) _____. | 1. _____ |
| 2. Energy that is released during a chemical change is called _____. | 2. _____ |
| 3. The amount of heat that is released or absorbed during a chemical change is called _____. | 3. _____ |
| 4. The amount of heat that is released or absorbed during a chemical change is called _____. | 4. _____ |
| 5. The amount of heat that is released or absorbed during a chemical change is called _____. | 5. _____ |
| 6. The amount of heat that is released or absorbed during a chemical change is called _____. | 6. _____ |
| 7. The amount of heat that is released or absorbed during a chemical change is called _____. | 7. _____ |
| 8. The amount of heat that is released or absorbed during a chemical change is called _____. | 8. _____ |
| 9. The amount of heat that is released or absorbed during a chemical change is called _____. | 9. _____ |
| 10. The amount of heat that is released or absorbed during a chemical change is called _____. | 10. _____ |
| 11. The amount of heat that is released or absorbed during a chemical change is called _____. | 11. _____ |
| 12. The amount of heat that is released or absorbed during a chemical change is called _____. | 12. _____ |
| 13. The amount of heat that is released or absorbed during a chemical change is called _____. | 13. _____ |
| 14. The amount of heat that is released or absorbed during a chemical change is called _____. | 14. _____ |
| 15. The amount of heat that is released or absorbed during a chemical change is called _____. | 15. _____ |
| 16. The amount of heat that is released or absorbed during a chemical change is called _____. | 16. _____ |
| 17. The amount of heat that is released or absorbed during a chemical change is called _____. | 17. _____ |
| 18. The amount of heat that is released or absorbed during a chemical change is called _____. | 18. _____ |
| 19. The amount of heat that is released or absorbed during a chemical change is called _____. | 19. _____ |
| 20. The amount of heat that is released or absorbed during a chemical change is called _____. | 20. _____ |

Part B True-False

Write true or false in the space provided. If false, write the correct word or phrase in the space provided.

1. The amount of heat that is released or absorbed during a chemical change is called enthalpy.
2. The amount of heat that is released or absorbed during a chemical change is called enthalpy.
3. The amount of heat that is released or absorbed during a chemical change is called enthalpy.

Section Review 11describing Chemical Reactions

Thomas Griffiths



Section Review 11describing Chemical Reactions:

The Enigmatic Realm of **Section Review 11describing Chemical Reactions**: Unleashing the Language is Inner Magic

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