

GUIDED PRACTICE PROBLEM 34 (page 307)

34. Calculate the percent composition of these compounds.

- a. ethane (C_2H_6)
- b. sodium bisulfate (NaHSO_4)

a. ethane (C_2H_6)**Analyze****Step 1.** List the knowns and the unknowns.**Knowns**mass of C in one mole ethane = $2 \times 12.0 \text{ g} = 24.0 \text{ g}$ mass of H in one mole ethane = $6 \times 1.0 \text{ g} = 6.0 \text{ g}$ molar mass of $\text{C}_2\text{H}_6 = 24.0 \text{ g} + 6.0 \text{ g} = 30.0 \text{ g}$ **Unknowns**

percent C = ? %

percent H = ? %

Because no masses are given, the percent composition can be determined based on the molar mass of the substance. The percent of an element in a compound is the mass of the element in the compound divided by the mass of the compound. To express the ratio as a percent, the ratio is multiplied by 100%.

Calculate**Step 2.** Solve for the unknown.

$$\text{percent C} = \frac{24.0 \text{ g C}}{30.0 \text{ g compound}} \times 100\% = 80.0\% \text{ C}$$

$$\text{percent H} = \frac{6.0 \text{ g H}}{30.0 \text{ g compound}} \times 100\% = 20.0\% \text{ H}$$

Evaluate**Step 3.** Does the result make sense?

The percents of the elements of the compound add up to 100%.

$$80.0\% + 20.0\% = 100\%$$

Section Chemical Quantities Practice Problem Answers

CO Houle



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The Practice of Chemistry Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. *The Practice of Chemistry* First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes *The Practice of Chemistry* the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H. Langley, 2014-04-14 Practice makes perfect and helps deepen your understanding of chemistry. Every high school requires a course in chemistry, and many universities require the course for majors in medicine, engineering, biology, and various other sciences. *1001 Chemistry Practice Problems For Dummies* provides students of this popular course the chance to practice what they learn in class, deepening their understanding of the material and allowing for supplemental explanation of difficult topics. *1001 Chemistry Practice Problems For Dummies* takes you beyond the instruction and guidance offered in *Chemistry For Dummies*, giving you 1,001 opportunities to practice solving problems from the major topics in chemistry. Plus, an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go. Gives you a chance to practice and reinforce the skills you learn in chemistry class. Helps you refine your understanding of chemistry. Practice problems with answer explanations that detail every step of every problem. Whether you're studying chemistry at the high school, college, or graduate level, the practice problems in *1001 Chemistry Practice Problems For Dummies* range in areas of difficulty and style, providing you with the practice help you need to score high at exam time.

Mapping College Chemistry Stephen DeMeo, 2019-03-01 This text is a chemistry problem-solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first-year college General Chemistry course. The book incorporates a chemistry problem-solving plan, one that uses an innovative graphic organizer strategy. The strategy successfully evaluated with students combines problem-solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry. Topical problem-solving will focus on limiting reactant stoichiometry, identifying types of chemical reactions, equilibrium, acid-base equilibria, and electrochemistry. Why would this resource be of interest to chemistry students? To be successful to get into a well-known college, medical school, physical therapy, or graduate program often requires that students get an A in your pre-requisite Introductory General Chemistry course. To make matters worse, many college professors feel that only a few students should get A grades, and therefore they give difficult exams that many students fail. This is the weeding out process that every pre-health student is apprehensive about. To succeed in this

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understanding of the basics of chemistry An integrated approach is employed in which related general chemistry organic chemistry and biochemistry topics are presented in adjacent chapters This approach helps students see the strong connections that exist between these three branches of chemistry and allows instructors to discuss these interrelationships while the material is still fresh in students minds

NTA JEE Main Chapter-wise DPP Sheets (25 Questions Pattern) for Chemistry 2nd Edition Disha Experts,2019-09-25

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Advances in Computer Assisted Learning P. R. Smith,2014-05-23 Advances in Computer Assisted Learning contains selected proceedings from the CAL Symposium on Computer Assisted Learning held at the University of Nottingham in the UK in 1985 This book reviews advances in computer assisted learning in the areas of curriculum development visually handicapped and disabled students project work in schools television viewdata and video applications database applications and engineering education and training This monograph has 35 chapters and opens with a discussion on the computing aspects of interactive video focusing on the design and production of the software used to control the videodisc developed by the Open University in the UK The next chapter illustrates a variety of case studies whereby local viewdata has been exploited by both teachers and their pupils in different parts of Europe Attention then turns to the use of computer assisted communication in the education of the visually impaired the use of microcomputers in teaching electronics and theoretical considerations in selecting software for language arts This text will be of interest to educators and policymakers who want to implement computer technology in the classroom

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Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text Schaum's Outline of

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