

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Section 6 1 Review Chemical Bonding Answer Key

Lauren Gardner



Section 6 1 Review Chemical Bonding Answer Key:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Section 6 1 Review Chemical Bonding Answer Key**

In a global inundated with screens and the cacophony of quick interaction, the profound energy and mental resonance of verbal art frequently disappear into obscurity, eclipsed by the regular assault of noise and distractions. Yet, nestled within the musical pages of **Section 6 1 Review Chemical Bonding Answer Key**, a fascinating perform of literary beauty that impulses with natural thoughts, lies an unforgettable trip waiting to be embarked upon. Composed by a virtuoso wordsmith, that mesmerizing opus manuals readers on a mental odyssey, softly revealing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse of the evocative examination, we shall embark upon an introspective exploration of the book is main styles, dissect their interesting writing type, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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