

Chapter 4 Atomic Structure

Section 4.2 The Structure of an Atom**(pages 108–112)**

This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.

Reading Strategy (page 108)

Monitoring Your Understanding Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Know About Atoms	What I Would Like to Learn	What I Have Learned
Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

Properties of Subatomic Particles (pages 108–109)

- What are three subatomic particles?
a. Protons b. Electrons c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.
a. nucleus **(b)** proton
c. neutron d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.

Comparing Subatomic Particles (pages 109–110)

- Circle the letters of properties that vary among subatomic particles.
a. color **(b)** mass
(c) charge **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.
(a) mass of 1 neutron = mass of 1 proton
b. mass of 2000 neutrons = mass of 1 proton
c. mass of 1 electron = mass of 1 proton
d. mass of 1 neutron = mass of 1 electron

Section Studying Atoms Answers

R Sanford



Section Studying Atoms Answers:

Enjoying the Song of Appearance: An Psychological Symphony within **Section Studying Atoms Answers**

In some sort of consumed by screens and the ceaseless chatter of instant communication, the melodic splendor and psychological symphony created by the prepared term usually fade into the back ground, eclipsed by the constant sound and interruptions that permeate our lives. But, situated within the pages of **Section Studying Atoms Answers** a wonderful literary treasure full of raw feelings, lies an immersive symphony waiting to be embraced. Crafted by an outstanding musician of language, that fascinating masterpiece conducts visitors on an emotional trip, skillfully unraveling the concealed melodies and profound affect resonating within each cautiously constructed phrase. Within the depths of the emotional review, we can explore the book is main harmonies, analyze its enthralling writing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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