

Balancing Chemical Equations Worksheet

1. 2 H₂ + 1 O₂ → 2 H₂O
2. 1 N₂ + 3 H₂ → 2 NH₃
3. 1 S₈ + 12 O₂ → 8 SO₃
4. 2 N₂ + 1 O₂ → 2 N₂O
5. 2 HgO → 2 Hg + 1 O₂
6. 6 CO₂ + 6 H₂O → 1 C₆H₁₂O₆ + 6 O₂
7. 1 Zn + 2 HCl → 1 ZnCl₂ + 1 H₂
8. 1 SiCl₄ + 4 H₂O → 1 H₄SiO₄ + 4 HCl
9. 2 Na + 2 H₂O → 2 NaOH + 1 H₂
10. 2 H₃PO₄ → 1 H₄P₂O₇ + 1 H₂O
11. 1 C₁₀H₁₆ + 8 Cl₂ → 10 C + 16 HCl
12. 1 CO₂ + 2 NH₃ → 1 OC(NH₂)₂ + 1 H₂O
13. 4 Si₂H₃ + 17 O₂ → 8 SiO₂ + 6 H₂O₃
14. 2 Al(OH)₃ + 3 H₂SO₄ → 1 Al₂(SO₄)₃ + 6 H₂O
15. 4 Fe + 3 O₂ → 2 Fe₂O₃
16. 1 Fe₂(SO₄)₃ + 6 KOH → 3 K₂SO₄ + 2 Fe(OH)₃
17. 2 C₇H₆O₂ + 15 O₂ → 14 CO₂ + 6 H₂O
18. 1 H₂SO₄ + 8 HI → 1 H₂S + 4 I₂ + 4 H₂O
19. 4 FeS₂ + 11 O₂ → 2 Fe₂O₃ + 8 SO₂
20. 2 Al + 3 FeO → 1 Al₂O₃ + 3 Fe
21. 1 Fe₂O₃ + 3 H₂ → 2 Fe + 3 H₂O
22. 1 Na₂CO₃ + 2 HCl → 2 NaCl + 1 H₂O + 1 CO₂
23. 2 K + 1 Br₂ → 2 KBr
24. 1 C₇H₁₆ + 11 O₂ → 7 CO₂ + 8 H₂O
25. 1 P₄ + 5 O₂ → 2 P₂O₅

Section Reactions And Equations Answers

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EPCRA Section 313 Questions and Answers, 1999

Progress in Physical Organic Chemistry Robert W. Taft, 2009-09-17 Progress in Physical Organic Chemistry is dedicated to reviewing the latest investigations into organic chemistry that use quantitative and mathematical methods These reviews help readers understand the importance of individual discoveries and what they mean to the field as a whole Moreover the authors leading experts in their fields offer unique and thought provoking perspectives on the current state of the science and its future directions With so many new findings published in a broad range of journals Progress in Physical Organic Chemistry fills the need for a central resource that presents analyzes and contextualizes the major advances in the field The articles published in Progress in Physical Organic Chemistry are not only of interest to scientists working in physical organic chemistry but also scientists working in the many subdisciplines of chemistry in which physical organic chemistry approaches are now applied such as biochemistry pharmaceutical chemistry and materials and polymer science Among the topics explored in this series are reaction mechanisms reactive intermediates combinatorial strategies novel structures spectroscopy chemistry at interfaces stereochemistry conformational analysis quantum chemical studies structure reactivity relationships solvent isotope and solid state effects long lived charged sextet or open shell species magnetic non linear optical and conducting molecules and molecular recognition

Chemistry: Concepts and Problems Clifford C. Houk, Richard Post, 1996-03-09

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