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Chapter 7 Chemical Reactions Section 7.1 Describing Reactions

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Name _____ Class _____ Date _____ Chapter 7 Chemical Reactions Section 7.1
Describe Changes in Reactions

Chemical Reactions - Purdue University College of AC

Chemical reactions produce substances with properties that are different from the reactants.

Chemical reactions are described by chemical equations.

Example: The reaction between hydrogen and oxygen to form water is represented by the

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Chapter 9: Physical and Chemical Changes

9.1 Matter - Physical Changes Chapter 9: Physical and Chemical Changes

Matter is anything that has mass and takes up space. It can be a solid, liquid, or gas.

How is boiling a solid a chemical reaction - Answers - 202

Answers - 202

Boiling is a physical change, not a chemical reaction. It is a change in state, not a change in composition.

What happens to atoms in a chemical reaction?

What happens to atoms in a chemical reaction?

Chemical reactions involve atoms and molecules. Atoms are the building blocks of matter. They are not created or destroyed in a chemical reaction.

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