

Lesson 14 Chemical Reactions

3

Types of Chemical Reactions

BEFORE YOU READ

After you read this section, you should be able to answer these questions:

- What is a synthesis reaction?
- What is a decomposition reaction?
- What is a displacement reaction?

National Science
Education Standards
PS 1b

Organizing Chemical Changes

There are thousands of different chemical reactions. It would be impossible to remember them all. However, reactions can be put in groups based on what happens during the reaction.

Most reactions fall into one of four types: synthesis, decomposition, single-displacement, and double displacement. Each type has a pattern that shows how reactants become products. In this section, drawings of students at a dance will show how each reaction works.

STUDY TIP

Construct Make a table listing each type of reaction and the pattern that shows how reactants become products.

Synthesis Reactions

A **synthesis reaction** is a reaction in which two or more substances combine to form a new compound. The substances that react can be elements or compounds. An example of a synthesis reaction is shown in the figure below. Sodium and chlorine react to make sodium chloride, which you know as table salt. The synthesis reaction is modeled by two people becoming a dancing couple. 



Sodium reacts with chlorine to form sodium chloride in this synthesis reaction.

READING CHECK

1. Describe What is a synthesis reaction?

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