

Naming Ionic Compounds

Name KEY Per _____

- | | |
|--|----------------------------|
| 1. NaCl | <u>sodium chloride</u> |
| 2. Na ₂ S | <u>sodium sulfide</u> |
| 3. B(CO ₃) ₃ | <u>boron carbonate</u> |
| 4. Mg(OH) ₂ | <u>magnesium hydroxide</u> |
| 5. MgF ₂ | <u>magnesium fluoride</u> |
| 6. (NH ₄) ₂ SO ₄ | <u>ammonium sulfate</u> |
| 7. K ₃ N | <u>potassium nitride</u> |
| 8. LiNO ₃ | <u>lithium nitrate</u> |
| 9. BaSO ₄ | <u>barium sulfate</u> |
| 10. Mg(NO ₃) ₂ | <u>magnesium nitrate</u> |
| 11. AgCl | <u>silver chloride</u> |
| 12. Al(OH) ₃ | <u>aluminum hydroxide</u> |
| 13. CaSO ₄ | <u>calcium sulfate</u> |
| 14. Ca ₃ P ₂ | <u>calcium phosphide</u> |
| 15. SrS | <u>strontium sulfide</u> |
| 16. BeCl ₂ | <u>beryllium chloride</u> |
| 17. NaI | <u>sodium iodide</u> |
| 18. AlCl ₃ | <u>aluminum chloride</u> |
| 19. MgCO ₃ | <u>magnesium carbonate</u> |

Section 3 Reinforcement Chemical Energy

RD Boyd

A decorative graphic consisting of a red and white circular shape, resembling a stylized 'C' or a partial circle, located to the right of the author's name.

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