Topics:

- 1. General Chemistry
- 2. Molecular Representations
- 3. Nomenclature
- 4. Alkanes/Cycloalkanes
- 5. Stereoisomers
- 6. Chemical reactivity/mechanism
- 7. SN/E Reactions

General Chemistry Questions

- 1. Identify all of the intermolecular forces present in each molecule:
 - a. CH₂OH
 - b. CH₂COCH₂CH₃
 - c. CH₂C(CH₃)₂
- 2. Fill out the blanks in the chart (VESPR Theory)

| Hybridization | Geometry | # of Bonds | # of Lone Pairs | Bond Angle |
|-----------------|-------------|--|-----------------|-------------------|
| | Tetrahedral | The state of the s | 0 | The second second |
| Sp ³ | | 3 | 1 | |
| | Bent | | | 109.5 |
| Sp² | 3,550 | 3 | 0 | |
| | | 2 | 1 | 120 |
| | | | 0 | 180 |

- 3. Determine if the following molecules have a dipole moment
 - a. CH₂CH₃
 - b. CH3CH2CH(NH₂)CH₃
 - c. CH₂Cl₂
 - d. H₂O*
 - e. HOOst

Molecular Representations Questions:

1. Write the condensed name for the following structures:



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