

## Polyatomic Ion Worksheet

Place the charges on each ion – Write the Formula – Name the Compound

Remember – The cation always goes first.

Example:	$\text{NH}_4^{+1}$	&	$\text{I}^{-1}$	Formula:	<u><math>\text{NH}_4\text{I}</math></u>	Name:	<u>Ammonium Iodide</u>
1.	$\text{NH}_4$	&	O	Formula:	_____	Name:	_____
2.	Ca	&	$\text{CO}_3$	Formula:	_____	Name:	_____
3.	$\text{HCO}_3$	&	Na	Formula:	_____	Name:	_____
4.	$\text{NO}_3$	&	Mg	Formula:	_____	Name:	_____
5.	K	&	$\text{SO}_4$	Formula:	_____	Name:	_____
6.	Li	&	$\text{PO}_4$	Formula:	_____	Name:	_____

## Finding the Formula from the Name

Example:	Lithium Chloride	Ions:	<u><math>\text{Li}^{+1}</math> &amp; <math>\text{Cl}^{-1}</math></u>	Formula:	<u><math>\text{LiCl}</math></u>
1.	Sodium Oxide	Ions:	_____	Formula:	_____
2.	Potassium Bromide	Ions:	_____	Formula:	_____
3.	Calcium Nitride	Ions:	_____	Formula:	_____
4.	Beryllium Sulfide	Ions:	_____	Formula:	_____
5.	Magnesium Iodide	Ions:	_____	Formula:	_____
6.	Aluminum Oxide	Ions:	_____	Formula:	_____
7.	Ammonium Bromide	Ions:	_____	Formula:	_____
8.	Calcium Bicarbonate	Ions:	_____	Formula:	_____
9.	Potassium Phosphate	Ions:	_____	Formula:	_____
10.	Aluminum Sulfate	Ions:	_____	Formula:	_____

# Polyatomic Ions Worksheet With Answers Physical Science If8767

**JA Banks**



## **Polyatomic Ions Worksheet With Answers Physical Science If8767:**

## Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Polyatomic Ions Worksheet With Answers Physical Science If8767**

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