

Periodic Table of the Elements

Normal boiling point is at 1°C
 ST = Triple Point
 Pressure is listed if not 1 atm
 Asterisk is listed if more than one allotrope

Atomic Number		Symbol		Name		Atomic Mass	
1	1	H	Hydrogen	1.008			
2	2	He	Helium	4.003			
3	3	Li	Lithium	6.941			
4	4	Be	Beryllium	9.012			
5	5	B	Boron	10.811			
6	6	C	Carbon	12.011			
7	7	N	Nitrogen	14.007			
8	8	O	Oxygen	15.999			
9	9	F	Fluorine	18.998			
10	10	Ne	Neon	20.180			
11	11	Na	Sodium	22.990			
12	12	Mg	Magnesium	24.305			
13	13	Al	Aluminum	26.982			
14	14	Si	Silicon	28.086			
15	15	P	Phosphorus	30.974			
16	16	S	Sulfur	32.06			
17	17	Cl	Chlorine	35.45			
18	18	Ar	Argon	39.948			
19	19	K	Potassium	39.098			
20	20	Ca	Calcium	40.078			
21	21	Sc	Scandium	44.956			
22	22	Ti	Titanium	47.88			
23	23	V	Vanadium	50.942			
24	24	Cr	Chromium	51.996			
25	25	Mn	Manganese	54.938			
26	26	Fe	Iron	55.845			
27	27	Co	Cobalt	58.933			
28	28	Ni	Nickel	58.69			
29	29	Cu	Copper	63.546			
30	30	Zn	Zinc	65.38			
31	31	Ga	Gallium	69.723			
32	32	Ge	Germanium	72.64			
33	33	As	Arsenic	74.922			
34	34	Se	Selenium	78.96			
35	35	Br	Bromine	79.904			
36	36	Kr	Krypton	83.80			
37	37	Rb	Rubidium	85.468			
38	38	Sr	Strontium	87.62			
39	39	Y	Yttrium	88.906			
40	40	Zr	Zirconium	91.224			
41	41	Nb	Niobium	92.906			
42	42	Mo	Molybdenum	95.94			
43	43	Tc	Technetium	98			
44	44	Ru	Ruthenium	101.07			
45	45	Rh	Rhodium	102.91			
46	46	Pd	Palladium	106.42			
47	47	Ag	Silver	107.87			
48	48	Cd	Cadmium	112.41			
49	49	In	Indium	114.82			
50	50	Sn	Tin	118.71			
51	51	Sb	Antimony	121.76			
52	52	Te	Tellurium	127.6			
53	53	I	Iodine	126.91			
54	54	Xe	Xenon	131.29			
55	55	Cs	Cesium	132.91			
56	56	Ba	Barium	137.33			
57	57	La	Lanthanum	138.91			
58	58	Ce	Cerium	140.12			
59	59	Pr	Praseodymium	140.91			
60	60	Nd	Niobium	144.24			
61	61	Pm	Promethium	145			
62	62	Sm	Samarium	150.36			
63	63	Eu	Europium	151.96			
64	64	Gd	Gadolinium	157.25			
65	65	Tb	Terbium	158.93			
66	66	Dy	Dysprosium	162.50			
67	67	Ho	Holmium	164.93			
68	68	Er	Erbium	167.26			
69	69	Tm	Thulium	168.93			
70	70	Yb	Ytterbium	173.05			
71	71	Lu	Lutetium	174.97			
72	72	Hf	Hafnium	178.49			
73	73	Ta	Tantalum	180.95			
74	74	W	Tungsten	183.84			
75	75	Re	Rhenium	186.21			
76	76	Os	Osmium	190.23			
77	77	Ir	Iridium	192.22			
78	78	Pt	Platinum	195.08			
79	79	Au	Gold	196.97			
80	80	Hg	Mercury	200.59			
81	81	Tl	Thallium	204.38			
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83	83	Bi	Bismuth	208.98			
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104	104	Rf	Rutherfordium	261			
105	105	Db	Dubnium	262			
106	106	Sg	Seaborgium	266			
107	107	Bh	Berkelium	264			
108	108	Hs	Hassium	277			
109	109	Mt	Moscovium	288			
110	110	Ds	Darmstadtium	285			
111	111	Rg	Roganium	289			
112	112	Cn	Copernicium	285			
113	113	Uut	Ununtrium	288			
114	114	Fl	Flerovium	289			
115	115	Uup	Ununpentium	288			
116	116	Lv	Livermorium	293			
117	117	Uus	Ununseptium	294			
118	118	Uuo	Ununoctium	294			

Lanthanide Series

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
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Actinide Series

Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
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Periodic Table Guide

David R. Klein



Periodic Table Guide:

The Periodic Table Tom Jackson, 2020-02-04 Which is the densest element Which has the largest atoms And why are some elements radioactive From the little known uses of gold in medicine to the development of the hydrogen bomb this is a fresh new look at the Periodic Table Combining cutting edge science with fascinating facts and stunning infographics this book looks at the extraordinary stories of discovery amazing properties and surprising uses of each elements whether solid liquid or gas naturally occurring synthesised or theoretical From hydrogen to oganesson this is a fact filled visual guide to each element each accompanied by technical data category atomic number weight boiling point as well as fun facts and stories about their discovery and surprising uses *The Periodic Table* Paul Parsons, 2014-03-11 As one of the most

recognizable images in science the periodic table is ingrained in our culture First drawn up in 1869 by Dmitri Mendeleev its 118 elements make up not only everything on our planet but also everything in the entire universe The Periodic Table looks at the fascinating story and surprising uses of each of those elements whether solid liquid or gas From the little known uses of gold in medicine to the development of the hydrogen bomb each entry is accompanied by technical data category atomic number weight boiling point presented in easy to read headers and a colour coding system that helps the reader to navigate through the different groups of elements A remarkable display of thought provoking science and beautiful photography this guide will allow the reader to discover the world afresh **Your Guide to the Periodic Table , Exploring the**

Elements Isabel Thomas, 2020 Science meets design in this comprehensive introduction to the chemical elements that make up our universe **A Kids' Guide to the Periodic Table** Edward P. Zovinka, Rose A. Clark, 2020-07-21 From aluminum to

zinc make it fun for kids 8 to 12 to discover all 118 elements on the periodic table Discover the building blocks of the entire world A Kids Guide to the Periodic Table takes you on an incredible journey through history and science that will teach you all about the 118 elements that make up well everything Go in depth with awesome profiles on each and every element that provide all their important elemental stats like their atomic number state group and more as well as awesome facts about the element and its discovery Take what you know about science and the world to a new level as you discover what makes the periodic table of elements so amazing A Kids Guide to the Periodic Table includes The periodic table explained Learn about the creation of the periodic table and get tons of info to help you understand the groups the order of elements and more Amazing discoveries Explore how elements like neon helium and californium were discovered as well as what they've helped scientists do Fun for you Find out how exciting science can be with an entertaining look into all the ways the elements affect your everyday life A fun fact filled science adventure awaits you with A Kids Guide to the Periodic Table **Chemistry**

Basics (Speedy Study Guide) Speedy Publishing, 2014-12-28 A study guide of chemistry basics would give students bite sized chunks of information on a complex subject making the task of learning it easier and more enjoyable Study guides present detailed information on a topic in a compact and simplistic way which helps the student grasp more of what they are

being taught Also the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory

Chemical Elements Pocket Guide Coventry House Publishing,2020-02-08 The Chemical Elements Pocket Guide serves as a portable reference for quick study and efficient review of the 118 elements on the periodic table This on the go resource details the physical and atomic properties of each element as well as their history and characteristics in bullet point format The book s small trim size 4 25 x 6 8 inches is intended to fit inside a lab coat pocket and the bound design means you no longer need to carry loose bulky flashcards that can be misplaced or destroyed Includes the updated names nihonium moscovium tennessine and oganesson for elements 113 115 117 and 118 respectively Information provided includes Atomic number Atomic symbol Element category Standard state Atomic mass Electron configuration Oxidation states Electronegativity Atomic radius Ionization energy Electron affinity Melting point Boiling point Density Year discovered Discovered by Appearance Natural occurrence Interesting fact

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills,Amina El-Ashmawy,2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein,2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

Periodic Table Basic Mark Jackson,2018

HowExpert Guide to Science HowExpert,2024-11-08 If you re ready to explore the world of science HowExpert Guide to Science is your essential resource covering everything from foundational principles to cutting edge innovations Designed for aspiring scientists students and curious minds this guide takes you on a journey through physics chemistry biology and beyond Focusing on key disciplines interdisciplinary connections ethics and practical knowledge this book offers insights tools and real world examples to truly understand science Chapter Highlights Introduction Discover science s role in everyday life and learn how to get the most from this guide Chapter 1 Foundations of Science Core principles the scientific method and the importance of hypotheses and theories Chapter 2 History of Science Major eras and groundbreaking discoveries that define modern science Chapter 3 Physics Matter energy mechanics thermodynamics and electromagnetism Chapter 4 Chemistry Matter s properties chemical reactions bonding and the periodic table Chapter 5 Biology Genetics evolution ecosystems and cellular biology Chapter 6 Earth Science Geology meteorology oceanography and environmental conservation Chapter 7 Astronomy Our solar system galaxies black holes and cosmology Chapter 8 Human Body Medical

Science Human anatomy disease prevention and medical advances Chapter 9 Environmental Science Human impact on the planet sustainability and climate change Chapter 10 Mathematics in Science Math's role in research from probability to calculus Chapter 11 Social Sciences Psychology sociology and the science of human behavior Chapter 12 Technology Applied Sciences Engineering computer science and technological innovation Chapter 13 Data Science Research Research design data analysis and statistical significance Chapter 14 Cutting Edge Sciences Advances in nanotechnology AI quantum mechanics and genetic engineering Chapter 15 Interdisciplinary Science The impact of cross field collaboration and interdisciplinary research Chapter 16 Practical Applications Everyday science in technology health and the environment Chapter 17 Scientific Literacy Skills to evaluate scientific information and avoid misconceptions Chapter 18 Citizen Science How to participate in public science initiatives and projects Chapter 19 Philosophy Ethics of Science Ethical considerations and the role of scientists in society Chapter 20 Science Communication Education Techniques for communicating science and inspiring future generations Chapter 21 Science Careers Explore career paths educational requirements and emerging roles in science Chapter 22 The Future of Science Emerging fields interdisciplinary research and future challenges Conclusion Summarizes key concepts inspires curiosity and reflects on science's impact Appendices Additional resources including a glossary recommended reading notable scientists hands on experiments and science events Why This Guide is Essential for Science Enthusiasts Comprehensive Learning Covers all major branches of science ideal for beginners and enthusiasts alike Practical Insights Real world applications examples and case studies deepen understanding Broad Perspective Connects scientific disciplines to reveal how science shapes our world Future Focused Highlights emerging fields and innovations offering a glimpse into science's future Start your journey with HowExpert Guide to Science and unlock a lifetime of discovery Whether deepening your knowledge or just beginning this guide opens doors to the wonders of science Dive in explore and let science reveal the mysteries of the universe HowExpert publishes how to guides on all topics from A to Z

Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise

Guinn, Rebecca Brewer, Rachel C. Lum, 2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text

Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With Organic Chemistry Student Study Guide and Solutions Manual 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

Periodic Table Advanced Mark Jackson, 2001-02-09 This 4 page laminated guide of the Periodic Table is a must for any Chemistry student

Elements Explorer, 1998

Mastering the Periodic Table Cybellium Ltd, 2024-10-26 Designed for professionals students and enthusiasts alike our comprehensive books empower you to stay ahead in a rapidly evolving digital world Expert Insights Our books provide deep actionable insights that bridge the gap between theory and practical application Up to Date Content Stay current with the latest advancements trends and best practices in IT AI Cybersecurity Business Economics and Science Each guide is regularly updated to reflect the newest developments and challenges Comprehensive Coverage Whether you re a beginner or an advanced learner Cybellium books cover a wide range of topics from foundational principles to specialized knowledge tailored to your level of expertise Become part of a global network of learners and professionals who trust Cybellium to guide their educational journey www.cybellium.com

The Elements Albert Stwertka, 2007-03-13 Students of chemistry and of life rejoice Put down your huge textbooks you re free This handy little guide will fit in your back pocket yet give volumes of detail about the elements enough to cover virtually any situation where you d need to refer to the periodic table This useful tool is an abridged version of the Oxford University Press book *A Guide to the Elements* which was honored with major awards from Science Books and Films magazine and the Society of School Librarians International It also won the Parents Choice Gold Award This tiny edition provides brief surveys of all 114 elements and features an easy to read complete periodic table in a quick foldout

A Guide to the Elements Albert Stwertka, 2012-02-01 Newly updated throughout and now covering 118 elements this crystal clear guide to the periodic table illuminates the basic concepts of chemistry as it traces the history and development of our knowledge of the material world In this fascinating volume Albert Stwertka makes complex ideas and terms easily understandable drawing upon engaging historical anecdotes and everyday examples to clarify the text which is complemented by numerous illustrations many in full color Since the second edition many new elements have been named and discovered including Darmstadtium Roentgenium and Copernicium and the elements currently called Ununtrium Ununpentium Ununhexium Ununseptium and Ununoctium The third edition provides thorough coverage of all these new discoveries In addition to the new elements Stwertka has brought the information about the elements in the second edition up to date based on the latest research He discusses a cylindrical molecule of carbon known as a nanotube which has become a do all wonder substance touted for use in everything from X ray machines to paint A new form of the element boron has been found that is nearly as hard as diamond Its superior heat resistance could make it attractive for certain industrial uses And a new particle detector using ultra pure liquid xenon has been constructed beneath 5 000 feet of rock in Italy to detect dark matter Stwertka also covers the 2010 Nobel winning work on graphene an ultrathin form of carbon that is vital for future generations of computers and touch screens the discovery of new superconductors and the development of new uses for the rare earth elements Bringing the periodic table into the 21st century this engrossing guide to the elements will fascinate everyone curious about the basic

building blocks of the material world *Survival Guide to General Chemistry* Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **OCR AS/A Level Chemistry A Student Guide: Modules 3 and 4** Mike Smith, 2015-11-30 Exam Board OCR Level AS A level Subject Chemistry First Teaching September 2015 First Exam Summer 2016 Written by experienced author Mike Smith this Student Guide for Chemistry Helps identify what you need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes

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