



Unit: Stoichiometry **Answer key** "Mole-Mole Calculations" - WS #1

Directions: For each of the following problems, show your balanced chemical equation and all of your calculations.

1. How many moles of potassium oxide (K_2O) will be formed when 1.52 moles of potassium reacts with oxygen according to the following reaction:



$$1.52 \text{ mol } K \times \frac{2 \text{ mol } K_2O}{4 \text{ mol } K} = 0.760 \text{ mol } K_2O$$

How many moles of copper are needed to react with sulfur to produce 0.25 moles of copper(I) sulfide according to the following reaction:



$$0.25 \text{ mol } Cu_2S \times \frac{2 \text{ mol } Cu}{1 \text{ mol } Cu_2S} = 0.50 \text{ mol } Cu$$

2. How many moles of sodium are needed to react with sulfuric acid to produce 3.75 moles of sodium sulfate according to the following (unbalanced) reaction:



$$3.75 \text{ mol } Na_2SO_4 \times \frac{2 \text{ mol } Na}{1 \text{ mol } Na_2SO_4} = 7.50 \text{ mol } Na$$

3. How many moles of calcium oxide will be produced when 1.6 moles of Iron(III)oxide reacts with calcium phosphate according to the following (unbalanced) reaction:



$$1.6 \text{ mol } Fe_2O_3 \times \frac{3 \text{ mol } CaO}{1 \text{ mol } Fe_2O_3} = 4.8 \text{ mol } CaO$$

4. How many moles of oxygen are produced when 5.12 moles of magnesium chlorate decomposes into magnesium chloride and oxygen?

Word equation: **magnesium chlorate $\rightarrow$ magnesium chloride + oxygen**

Formula equation: **$Mg(ClO_3)_2 \rightarrow MgCl_2 + 3O_2$**
(must be balanced!)

$$5.12 \text{ mol } Mg(ClO_3)_2 \times \frac{3 \text{ mol } O_2}{1 \text{ mol } Mg(ClO_3)_2} = 15.4 \text{ mol } O_2$$

5. How many moles of Chromium(III) nitrate are produced when Chromium reacts with 0.85 moles of Lead(IV) nitrate to produce chromium(III) nitrate and lead?

Word equation: **chromium + lead (IV) nitrate $\rightarrow$ chromium (III) nitrate + lead**

Formula equation: **$4Cr + 3Pb(NO_3)_4 \rightarrow 4Cr(NO_3)_3 + 3Pb$**
(must be balanced!)

$$0.85 \text{ mol } Pb(NO_3)_4 \times \frac{4 \text{ mol } Cr(NO_3)_3}{3 \text{ mol } Pb(NO_3)_4} = 1.1 \text{ mol } Cr(NO_3)_3$$

Section Quiz Introduction To Stoichiometry Answers

Judd E. Hollander



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Veenstra,Fernando F. Padró,Julie A. Furst-Bowe,2012-05-15 In July 2011 the ASQ Education Division held its first Advancing the STEM Science Technology Engineering and Mathematics Agenda in Education the Workplace and Society Conference at the University of Wisconsin Stout This publication is a selection of papers and workshops from this groundbreaking conference The ideas presented here will help other educators and policy makers to develop their own innovative high impact ideas for inspiring student interest in STEM careers improving the delivery of STEM education at their schools and colleges and helping STEM college graduates transition to the workplace The chapters in this book reflect research and best practices integrating the ideas of continuous improvement in combination with a can do attitude to provide a valuable resource that will lead others to consider similar innovative and collaborative educational structures that will drive more interest in STEM majors in college and provide for our next generation of scientists technicians and engineers Prior to reviewing Advancing the STEM Agenda I had a list in my mind of topics that I hoped would be addressed I m very pleased with how many are covered and covered well This project succeeds at the challenge of providing not only beneficial breadth but also important depth Because our public private partnership has been committed explicitly to continuous improvement for more than a decade I couldn t help but notice as the editors also point out in their conclusion the extent to which continuous improvement is a common thread throughout the book That speaks to the book s practical utility in many settings and on a long term basis No less valuable is the discussion of student motivation by many of the authors which STEM teachers in our area have identified as a major issue of interest to them in recent surveys Richard Bogovich Executive Director Rochester Area Math Science Partnership Minnesota Veenstra Padr and Furst Bowe provide a huge contribution to the field of STEM education We all know the statistics and of the huge need in the area of STEM students and education but what has been missing are application and success stories backed by research and modeling The editors have successfully contributed to our need by focusing on collaborative models building the K 12 pipeline showing what works at the collegiate level connecting across gender issues and illustrating workforce and innovative ideas John J Jasinski President Northwest Missouri State University Advancing the STEM Agenda provides a broad set of current perspectives that will contribute in many ways to advancing the understanding and enhancement of education in science education and engineering This work is packed with insights and perspectives from experienced educators and bridges the transition from education to workplace John Dew Senior Vice Chancellor Troy University General Catalog University of California, Los Angeles,1975 **Bulletin** University of Minnesota,1961 *General Catalogue* University of California, Los Angeles,1975 **Resources in Education** ,1980

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