

Stoichiometry

Quiz 11*9 Section 9

Section 9: Introduction to Stoichiometry

In the space provided, write the letter of the best response that best completes each problem or best answers each question.

1. Which of the following is the formula for a compound that does not contain an ionic bond and with the lowest molecular weight?
 - A. H_2O
 - B. CH_4
 - C. CO_2
 - D. NH_3
2. Which of the following is a correct statement?
 - A. The mass of a substance is proportional to the number of moles.
 - B. The mass of a substance is proportional to the number of molecules.
 - C. The mass of a substance is proportional to the number of atoms.
 - D. The mass of a substance is proportional to the number of ions.
3. Which of the following is a correct statement?
 - A. The mass of a substance is proportional to the number of moles.
 - B. The mass of a substance is proportional to the number of molecules.
 - C. The mass of a substance is proportional to the number of atoms.
 - D. The mass of a substance is proportional to the number of ions.
4. The chemical equation $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ is not balanced. Which of the following is the correct balanced equation?
 - A. $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 - B. $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$
 - C. $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow 2\text{CO}_2 + \text{H}_2\text{O}$
 - D. $\text{C}_2\text{H}_4 + \text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
5. Which of the following is a correct statement?
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 - B. The mass of a substance is proportional to the number of molecules.
 - C. The mass of a substance is proportional to the number of atoms.
 - D. The mass of a substance is proportional to the number of ions.

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