



COVALENT BONDING

Practice Problems

Do your answers to these very practice problems.

SECTION 8.1 MOLECULAR COMPOUNDS

1. Classify each of the following compounds as ionic or covalent.

- a. Na_2SO_4 b. CH_3NH_2 c. HCl d. H_2O e. H_2O_2 f. H_2S g. H_2SO_4 h. H_2SO_3 i. H_2SO_2 j. H_2SO k. H_2SO l. H_2SO m. H_2SO n. H_2SO o. H_2SO p. H_2SO q. H_2SO r. H_2SO s. H_2SO t. H_2SO u. H_2SO v. H_2SO w. H_2SO x. H_2SO y. H_2SO z. H_2SO

2. Write the formula for the following molecular compounds.

- a. H_2O b. H_2O_2 c. H_2O d. H_2O e. H_2O f. H_2O g. H_2O h. H_2O i. H_2O j. H_2O k. H_2O l. H_2O m. H_2O n. H_2O o. H_2O p. H_2O q. H_2O r. H_2O s. H_2O t. H_2O u. H_2O v. H_2O w. H_2O x. H_2O y. H_2O z. H_2O

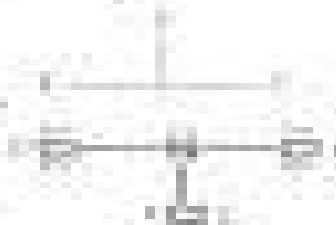
3. Write the formula for the following molecular compounds. (Note: The following are the names of the compounds.)
 a. H_2O b. H_2O_2 c. H_2O d. H_2O e. H_2O f. H_2O g. H_2O h. H_2O i. H_2O j. H_2O k. H_2O l. H_2O m. H_2O n. H_2O o. H_2O p. H_2O q. H_2O r. H_2O s. H_2O t. H_2O u. H_2O v. H_2O w. H_2O x. H_2O y. H_2O z. H_2O

SECTION 8.2 THE NATURE OF COVALENT BONDING

1. Draw the Lewis structure for the following molecule.



2. Draw the Lewis structure for the following molecule.



3. Draw the Lewis structure for the following molecule.



4. Draw the Lewis structure for the following molecule.

5. Draw the Lewis structure for the following molecule. (Note: The following are the names of the compounds.)



6. Draw the Lewis structure for the following molecule.

SECTION 8.3 BONDING: THE ORBITAL

1. Write the formula for the following molecular compounds. (Note: The following are the names of the compounds.)
 a. H_2O b. H_2O_2 c. H_2O d. H_2O e. H_2O f. H_2O g. H_2O h. H_2O i. H_2O j. H_2O k. H_2O l. H_2O m. H_2O n. H_2O o. H_2O p. H_2O q. H_2O r. H_2O s. H_2O t. H_2O u. H_2O v. H_2O w. H_2O x. H_2O y. H_2O z. H_2O

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RD Boyd



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