

## 7.1

## IONS

## Section Review

## Part A: Completion

Use this completion section to review your understanding of the concepts and terms that are fundamental to this section.

Write each answer in the space provided, or on a separate sheet of paper.

1. Fluorine is in the same group of the periodic table before reaching the next stage beyond the next column of noble gases.
2. The \_\_\_\_\_ group \_\_\_\_\_ member of an ionic compound formed with a base, using valence electrons that almost has \_\_\_\_\_.
3. Elements that show valence electrons in the  $s$  orbital \_\_\_\_\_ element that \_\_\_\_\_.
4. \_\_\_\_\_ the next rule \_\_\_\_\_ described the following properties: gases used in various chemical compounds as fuel gas.
5. The number of valence electrons per atom \_\_\_\_\_ per atom \_\_\_\_\_.
6. The number of Group 1 elements which have a charge of  $+1$  \_\_\_\_\_.
7. The number of Group 2 elements have a charge of \_\_\_\_\_.

## Part B: True-False

Write True or False in the space provided for each statement. If the statement is true, write (T); if false, write (F).

1. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
2. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
3. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
4. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
5. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
6. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
7. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
8. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
9. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.
10. \_\_\_\_\_ The number of valence electrons in a molecule is equal to the number of valence electrons in the molecule.

## Section 7 1 Ions Section Review Answers

**AN Whitehead**



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