

# Radioactivity

**Radioactivity** is the spontaneous breaking up of unstable nuclei accompanied by the emission of one or more types of radiation.

A **Geiger-Müller tube** is used to detect the presence of radiation.

## Types of Radiation

| Type            | Nature                                | Charge | Penetrating range    |
|-----------------|---------------------------------------|--------|----------------------|
| Alpha particles | 2 protons + 2 neutrons (a He nucleus) | +2     | a few cm of air      |
| Beta particles  | Electrons                             | -1     | 5mm Aluminium        |
| Gamma rays      | Electromagnetic radiation             | 0      | Thick lead, concrete |

## Differences between chemical reactions and nuclear reactions

| Nuclear Reaction               | Chemical Reaction             |
|--------------------------------|-------------------------------|
| Change occurs in the nucleus   | Involves electrons            |
| New element formed in reaction | No new element formed         |
| Nuclear radiation given off    | No nuclear radiation released |

# Section Radioactivity Answers Notes

**DP Hallahan**



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