

Chapter 2 Properties of Matter

Section 2.2 Physical Properties**(pages 45–51)**

This section discusses physical properties and physical changes. It also explains how physical properties can be used to identify materials, select materials, and separate mixtures.

Reading Strategy (page 45)

Building Vocabulary As you read, write a definition for each term in the table below. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Defining Physical Properties	
Physical Property	Definition
Viscosity	The tendency of a liquid to resist flowing
Malleability	The ability of a solid to be hammered without shattering
Melting Point	The temperature at which a solid changes to a liquid

Examples of Physical Properties (pages 45–47)

1. A physical property is any characteristic of a material that can be observed or measured without changing the composition of the substances in the material.
2. Explain why a wooden spoon is a better choice than a metal spoon for stirring a boiling pot of soup. The handle of a wooden spoon will stay cool because wood is not a good conductor of heat.
3. Is the following sentence true or false? A liquid with a high viscosity flows more slowly than a liquid with a low viscosity at the same temperature. true
4. Is the following sentence true or false? Discovering which of two materials can scratch the other is a way to compare the hardness of the materials. true

Match each term to its definition.

Term	Definition
<u>c</u> 5. viscosity	a. The ability of a solid to be hammered without shattering
<u>d</u> 6. conductivity	b. The temperature at which a substance changes from a liquid to a gas
<u>a</u> 7. malleability	c. The resistance of a liquid to flowing
<u>f</u> 8. melting point	d. The ability to allow heat to flow
<u>b</u> 9. boiling point	e. The ratio of the mass of a substance to its volume
<u>e</u> 10. density	f. The temperature at which a substance changes from a solid to a liquid

Section Physical Properties Of Solutions Answer Key

Sir Norman Lockyer



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Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

Selected Physical Properties of Solutions of Calcium in Liquid Ammonia Wa She Wong, 1966

Solvents and Solutions: Structure and Properties Keshra Sangwal, 2021-08-06 A UNIQUE BOOK ON THE PRESENT STATUS OF SOLVENTS AND SOLUTIONS WITH IMPORTANT PROBLEMS RELATED TO THEIR STRUCTURE AND PROPERTIES. The literature on the properties of solvents and solutions used in academic research and in a wide range of industries has grown enormously during the last four decades and is scattered in different specialized journals. *Solvents and Solutions* is a groundbreaking text that offers a systematic compilation of important problems related to selected properties of solvents and solutions based on the literature published so far. The author places emphasis on explaining the basic concepts involved in understanding the properties and behavior of various solvents and solutions of electrolytes and nonelectrolytes in a consistent manner. After a description of the general characteristics of structure of solvents and solutions and the solubility of electrolytes and nonelectrolytes under normal temperature and pressure conditions, the book first deals with different aspects of the density and the refractive index of solvents and dilute as well as concentrated solutions and finally with the transport i.e. viscosity and electric conductivity and thermal properties of solvents and solutions. *Solvents and Solutions* is the first text devoted to the description and discussion of their properties since the publication of a monograph on the physical properties of aqueous electrolyte solutions more than three decades ago. The main features of this book are: Reflects developments in the investigation of solvents and solutions during the last three decades. Outlines basic concepts involved in understanding the properties and behavior of solvents and solutions. Describes and discusses different properties of ionic liquids as solvents and the behavior of their mixtures with other commonly used solvents. Contents of different chapters are not only self-contained but the contents are practically independent of each other. Written as a practical guide for researchers who are looking for an up-to-date overview of the physical and transport properties of solvents and solutions and as a reference source for workers in chemical industries and related fields and for graduate students of chemical engineering and physical chemistry.

Biomedical Index to PHS-supported Research, 1991 Research Awards Index, 1989 **Medical**

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