



Percentage Yield

Sample Problem H Solution, *continued*



Theoretical yield

$$\begin{aligned} 36.8 \text{ g C}_6\text{H}_6 &\times \frac{1 \text{ mol C}_6\text{H}_6}{78.12 \text{ g C}_6\text{H}_6} \times \frac{1 \text{ mol C}_6\text{H}_5\text{Cl}}{1 \text{ mol C}_6\text{H}_6} \times \frac{112.56 \text{ g C}_6\text{H}_5\text{Cl}}{1 \text{ mol C}_6\text{H}_5\text{Cl}} \\ &= 53.0 \text{ g C}_6\text{H}_5\text{Cl} \end{aligned}$$

Percentage yield

$$\text{percentage yield C}_6\text{H}_5\text{Cl} = \frac{\text{actual yield}}{\text{theoretical yield}} \times 100$$

$$\text{percentage yield} = \frac{38.8 \text{ g}}{53.0 \text{ g}} \times 100 = 73.2\%$$



Section Limiting Reactants And Percentage Yield

Leo J. Malone, Theodore O. Dolter



Section Limiting Reactants And Percentage Yield:

The Enigmatic Realm of **Section Limiting Reactants And Percentage Yield**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Section Limiting Reactants And Percentage Yield** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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