

Chapter 5

Section 2 Electron Configuration and the Periodic Table

Periodic Table of the Elements

Periodic Table of Elements

Groups:

- Group 1: Alkali metals
- Group 2: Alkaline earth metals
- Groups 3-10: Transition metals
- Groups 11-12: Other metals
- Groups 13-18: Other nonmetals

Periods:

- Period 1: Hydrogen
- Period 2: Helium
- Period 3: Lithium, Beryllium, Boron, Carbon, Nitrogen, Oxygen, Fluorine, Neon
- Period 4: Potassium, Calcium, Scandium, Titanium, Vanadium, Chromium, Manganese, Iron, Cobalt, Nickel, Copper, Zinc, Gallium, Germanium, Arsenic, Selenium, Bromine, Krypton
- Period 5: Rubidium, Strontium, Yttrium, Zirconium, Niobium, Molybdenum, Technetium, Ruthenium, Rhodium, Palladium, Silver, Cadmium, Indium, Tin, Antimony, Tellurium, Iodine, Xenon
- Period 6: Cesium, Barium, Lanthanum, Cerium, Praseodymium, Neodymium, Promethium, Samarium, Europium, Gadolinium, Terbium, Dysprosium, Holmium, Erbium, Thulium, Ytterbium, Lutetium
- Period 7: Francium, Radium, Actinium, Thorium, Protactinium, Uranium, Neptunium, Plutonium, Americium, Curium, Bk, Cf, Es, Fm, Md, No, Lr

Notes:

- Elements in the same group have similar chemical properties.
- Elements in the same period have the same number of electron shells.
- The periodic table is a model of the universe's most stable and most common matter.

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JS Bruner



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