

Radioactive Decay Worksheet #2

Isotopes

- For atoms of the two different isotopes Lithium-6 and Lithium-7...
 - ...they must have the same number of what type of subatomic particle?
 - ...they must have different numbers of what type of subatomic particle?
- For each of the following isotopes, write in any missing information about that particular isotope.

Name	Oxygen-15	$^{58}_{28}\text{Ni}$	Zirconium- ____	Argon- ____	____ - ____	____ - ____
# of protons					9	30
# of neutrons			50	22	10	34

- Potassium (K) has three naturally-occurring isotopes, Potassium-39, Potassium-40, and Potassium-41. Using a periodic table, which isotope do you think is the most abundant in nature?

What leads you to this conclusion?

Decay Reactions

Fill in the blank in each of the following decay reactions with the correct decay particle or decayed nucleus that will balance the decay reaction, and also state whether it is alpha or beta decay.



- Write out the decay reaction for a nucleus of Lead-210 decaying by alpha decay.

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