BOARD QUESTION PAPER: MARCH 2022 CHEMISTRY

Time: 3 Hrs Max. Marks: 70

General Instructions:

The question paper is divided into four sections.

	(1)	Section A: Q. No. I contains Ten multiple choice type of questions carrying One mark each. O. No. 2 contains Eight very short answer type of questions carrying One mark each.	
	(2)	Section B: Q. No. 3 to Q. No. 14 are Twelve short answer type of questions carrying Two marks each. (Attempt any Eight).	
	(3)	Section C: Q. No. 15 to Q. No. 26 are Twelve short answer type of questions carrying Three marks each. (Attempt any Eight).	
	(4)	Section D: Q. No. 27 to Q. No. 31 are Five long answer type of questions carrying Four marks each. (Attempt any Three).	
	(5)	Use of log table is allowed. Use of calculator is not allowed.	
	(6)	Figures to the right indicate full marks.	
	(7)	For each multiple choice type of question, it is mandatory to write the correct answer along with its	
		alphabet. e.g. (a)/(b)/(c)/(d) etc.	
		narks(s) shall be given, if ONLY the correct answer or the alphabet of the correct answer is written.	
	Only the first attempt will be considered for evaluation.		
		SECTION - A	
Q.1.	Select and write the correct answer for the following multiple choice type of questions: [10]		
	(i)	The co-ordination number of atoms in body centred cubic structure (bcc) is	
		(a) 4 (b) 6 (c) 8 (d) 12	
	(iii)	In calculating osmotic pressure, the concentration of solute is expressed in	
		(a) molarity (b) molality	
		(c) mole fraction (d) percentage mass	
	(iiii)	The enthalpy change for the chemical reaction $H_2O_{(s)} \longrightarrow H_2O_{(t)}$ is called enthalpy of	
		(a) vapourisation (b) fusion	
		(c) combustion (d) sublimation	
	(iv)	Which of the following transition element shows maximum oxidation state?	
	533.5	(a) Sc (b) Fe (c) Mn (d) V	
	(v)	The correct formula for the complex compound, sodium hexacyanoferrate (III) is	
	20004	(a) Na [Fe(CN) ₆] (b) Na ₂ [Fe(CN) ₆]	
		(c) Na ₃ [Fe(CN) ₆] (d) Na ₄ [Fe(CN) ₆]	

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Table of Contents Oct 2014chemistry Q Paper Maharashtra

- 1. Understanding the eBook Oct 2014chemistry Q Paper Maharashtra
 - The Rise of Digital Reading Oct 2014chemistry Q Paper Maharashtra
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Oct 2014chemistry Q Paper Maharashtra
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Oct 2014chemistry Q Paper Maharashtra
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Oct 2014chemistry Q Paper Maharashtra
 - Personalized Recommendations
 - o Oct 2014chemistry Q Paper Maharashtra User Reviews and Ratings
 - o Oct 2014chemistry Q Paper Maharashtra and Bestseller Lists
- 5. Accessing Oct 2014chemistry Q Paper Maharashtra Free and Paid eBooks
 - Oct 2014chemistry Q Paper Maharashtra Public Domain eBooks
 - o Oct 2014chemistry Q Paper Maharashtra eBook Subscription Services
 - Oct 2014chemistry Q Paper Maharashtra Budget-Friendly Options
- 6. Navigating Oct 2014chemistry Q Paper Maharashtra eBook Formats
 - o ePub, PDF, MOBI, and More
 - o Oct 2014chemistry Q Paper Maharashtra Compatibility with Devices
 - Oct 2014chemistry Q Paper Maharashtra Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Oct 2014chemistry Q Paper Maharashtra
 - o Highlighting and Note-Taking Oct 2014chemistry Q Paper Maharashtra
 - o Interactive Elements Oct 2014chemistry Q Paper Maharashtra
- 8. Staying Engaged with Oct 2014chemistry Q Paper Maharashtra

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Oct 2014chemistry Q Paper Maharashtra
- 9. Balancing eBooks and Physical Books Oct 2014chemistry Q Paper Maharashtra
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Oct 2014chemistry Q Paper Maharashtra
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Oct 2014chemistry Q Paper Maharashtra
 - Setting Reading Goals Oct 2014chemistry Q Paper Maharashtra
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Oct 2014chemistry Q Paper Maharashtra
 - o Fact-Checking eBook Content of Oct 2014chemistry Q Paper Maharashtra
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - o Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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