

Half Life and Radioactive Decay

Name: _____

Instructions: The *half-life* of a radioactive substance is a characteristic constant. It measures the time it takes for a given amount of the substance to become reduced by half as a consequence of decay, and therefore, the emission of radiation. Scientists use half-life to calculate amount of radioactive isotopes present in a sample.

Sample Half-life Problem: A 100 gram sample of carbon-14 undergoes radioactive decay until 12.5 grams of carbon-14 remains in the original sample. This process took 17,190 years. How many half-lives have passed? What is the half-life of ^{14}C ?

Step 1: Make a data table with the original isotope, half-lives and time passed as its column headers.

Isotope (g)	Number of Half-lives Passed	Time Passed (Years)
100g	0	0
50	1	
25	2	

Step 2: Using the half-life principle, deduct the correct amount of isotope and record the number of half-lives passed until you reach the given number of years.

Amount of Isotope (g)	Number of Half-lives Passed	Time Passed (Years)
100g	0	

Step 3: Divide the number of year passed by the number of half-lives passed: $17,190/9 = 5730$ years in one half-life. Fill in the remainder of the chart.

Amount of Isotope (g)	Number of Half-lives Passed	Time Passed (Years)
100g	0	0
50	1	5730
25	2	11460
12.5	3	17190

Half-life Problems:

- What is the half-life of ^{235}U if a 500g sample decays to 30.25 grams in 63 hours?
- How much of a 100 g sample of ^{14}C would remain after 37 years if the half-life is 12.35 years?
- The half-life of ^{239}Pu is 1.4×10^4 years. If there are 6.25 g of the sample left after 2.8×10^4 years, how many grams were in the original sample?
- The half-life of Iodine-131 is 8.04 days. How long will it take a 200 g sample to decay into 6.25g?

Radioactive Decay And Half Life Worksheet Answers

Holt Rinehart & Winston



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Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

Radiologic Science for Technologists Stewart C. Bushong, 1997 The purpose of this textbook is to convey a working knowledge of radiologic physics and to prepare radiography students for the certification exam by the ARRT The textbook also provides a standard of knowledge from which practicing radiographers can make decisions about technical factors and diagnostic image quality in the work place This edition gives an expanded coverage of quality management which includes all of the content on the ARRT It also includes coverage of new cardiovascular interventional equipment and recent advances in spiral CT and digital radiography Keeps students informed and up to date with respect to professional standards and requirements

Nuclear Energy, 1985 **Addison-Wesley Science Insights**, 1996 Prentice Hall Science Explorer Michael J. Padilla, 2002 **Laboratory Manual in Physical Geology** American

Geological Institute, 1997 This Laboratory Manual in Physical Geology is a richly illustrated user friendly laboratory manual for teaching introductory geology and geoscience Cutnell & Johnson Physics John D. Cutnell, David Young, Kenneth W.

Johnson, Shane Stadler, 2022 The newly revised Twelfth Edition of Cutnell's Physics delivers an effective and accessible introduction to college and university physics It contains easy to follow explanations of critical math and problem solving concepts From kinematics to work and energy temperature heat electricity magnetism and optics as well as foundational concepts in more advanced subjects like special relativity Physics is the ideal introductory text for students from any background The greatest strength of the text is the synergistic relationship it develops between problem solving and conceptual understanding The book lays emphasis on building relevance of physics in day to day living and highlights the physics principles that come into play A wide range of applications that are biomedical in nature and others that deal with modern technology

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