

Republic of the Philippines

Department of Education REGION III

REGIONAL DIAGNOSTIC ASSESSMENT IN PHYSICAL SCIENCE

Name	Score
Direction: Choose the correct answer for	each item.
What type of nuclear synthesis is belief	eved to be the origin of the production of heavy elements?
A. Big Bang Nucleosynthesis	C. Stellar Nucleosynthesis
B. Primordial Nucleosynthesis	D. Supernova Nucleosynthesis
2. What is the ability of an atom in a com	pound to attract electrons towards itself called?
A. Electromotive force	C. Electronegativity
B. Electron Affinity	D. Ionization energy
3. Which of the following is the measure	of a liquid's resistance to flow?
A. Boiling point	C. Solubility
B. Melting point	D. Viscosity
4. Which of the following is the building b	block of protein?
A. Amino acid	C. Monomer
B. Glycogen	D. Nucleotide
5. What do you call the substance the consumed by the reaction?	nat increases the rate of chemical reaction without itself bein
A. Activation energy	C. Product
B. Catalyst	D. Reactant
6. Which of the following describes a stel A. It is the process by which elements B. It is the formation of elements duri C. It is the formation of light elements D. It is the process by which element	s are formed within stars ng a supernova explosion s such as helium and hydrogen
7. Which of the following statements is T	RUE about non polar molecules?
A. They have high boiling point	
 B. They have high melting point 	
 C. They have low surface tension 	
 They have low vapor pressure 	
8. Which will result in an ion-dipole inter-	action?
 A. Attraction between an ion and a po 	
 B. Attraction between a polar with a p 	
 C. Repulsion between dipole and and 	
 D. Repulsion between a polar and a r 	nonpolar molecule
9. Why are lipids insoluble in water?	
 A. Because Lipids are hydrophilic 	
 B. Because Lipids are hydrophobic 	
 C. Because Lipids are neutral D. Because Lipids are ionic compour 	nd .
10. What will be the effect of temperature	
 A. It increases molecular concentration 	
B. It increases the number of effective	e collisions among reactants, thus increasing the reaction rate

C. It decreases the system's energy and limits how much further reaction can take place
 D. It decreases the number of collisions and the energy of molecules to limit further reaction

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