

Chemistry Unit Test #5 SNC2D Name:

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Assessment of Learning

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Multiple Choice: Circle the most correct answer. (15 P)

1. The symbol for the element potassium is ____.

- A) K B) P C) K D) S E) Ca

2. Which among the following is NOT both a molecule and a compound?

- A) CaH_2O_4 B) H_2O C) CO_2 D) NaCl

3. Which of the following statements concerning differences between ionic and covalent compounds is correct?

- A) Covalent compounds contain 2 elements, and ionic compounds contain 3 or more elements.
B) Ionic compounds contain oxygen, and covalent compounds do not.
C) Covalent compounds form molecules, and ionic compounds do not.
D) more than one correct response.
E) no correct response.

4. Which of the following compounds would you expect to be ionic?

- A) SF_6 B) H_2O C) H_2O_2
D) NH_3 E) CaO

5. Which chemical group is called the alkali metals?

- A) group 1 B) group 2 C) groups 3-12
D) group 17 E) group 18

6. The number of protons in an atom of ^{65}Zn is:

- A) 65 B) 35 C) 30 D) 1.

7. An atom of atomic number 53 and mass number 127 contains how many neutrons?

- A) 53 B) 74 C) 127 D) 180

8. Most of the elements on the periodic table are _____.

- A) gases B) nonmetals C) metalloids D) liquids E) metals

9. Which one of the following is NOT true about the alkali metals?

- A) They are low density solids at room temperature.
B) They all readily form ions with a +1 charge.

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S Marginson



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Table of Contents Physical Science Chemistry Test Answers Unit 2

1. Understanding the eBook Physical Science Chemistry Test Answers Unit 2
 - The Rise of Digital Reading Physical Science Chemistry Test Answers Unit 2
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Chemistry Test Answers Unit 2
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Chemistry Test Answers Unit 2
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Chemistry Test Answers Unit 2
 - Personalized Recommendations
 - Physical Science Chemistry Test Answers Unit 2 User Reviews and Ratings
 - Physical Science Chemistry Test Answers Unit 2 and Bestseller Lists

5. Accessing Physical Science Chemistry Test Answers Unit 2 Free and Paid eBooks
 - Physical Science Chemistry Test Answers Unit 2 Public Domain eBooks
 - Physical Science Chemistry Test Answers Unit 2 eBook Subscription Services
 - Physical Science Chemistry Test Answers Unit 2 Budget-Friendly Options
6. Navigating Physical Science Chemistry Test Answers Unit 2 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Chemistry Test Answers Unit 2 Compatibility with Devices
 - Physical Science Chemistry Test Answers Unit 2 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Chemistry Test Answers Unit 2
 - Highlighting and Note-Taking Physical Science Chemistry Test Answers Unit 2
 - Interactive Elements Physical Science Chemistry Test Answers Unit 2
8. Staying Engaged with Physical Science Chemistry Test Answers Unit 2
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Chemistry Test Answers Unit 2
9. Balancing eBooks and Physical Books Physical Science Chemistry Test Answers Unit 2
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Chemistry Test Answers Unit 2
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Chemistry Test Answers Unit 2
 - Setting Reading Goals Physical Science Chemistry Test Answers Unit 2
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Chemistry Test Answers Unit 2
 - Fact-Checking eBook Content of Physical Science Chemistry Test Answers Unit 2
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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