

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

1								18
X	2		13	14	15	16	17	L
U	R		A	T	H P	G	Q	W
J	M		A I	D P	Z	E	K	V

- Clue #1 U and J are alkali metals. J has more energy levels than U.
- Clue #2 I has 4 valence electrons on the ~~3rd~~^{2nd} energy level.
- Clue #3 M is a metal in period 3 with 2 valence electrons
- Clue #4 X has one proton in its nucleus
- Clue #5 Q has 2 energy levels and is a halogen
- Clue #6 L is the only noble gas that doesn't have 8 valence electrons
- Clue #7 A is the only metalloid in the 13th group.
- Clue #8 I has 3 valence electrons on the third energy level
- Clue #9 V and W are un-reactive elements. V has more electrons than W.
- Clue #10 E and G have a common ion of -2 (E^{-2} , G^{-2}). E has 16 neutrons.
- Clue #11 R has an nuclear symbol of 3_1R
- Clue #12 K has one less electron than V
- Clue #13 D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.
- Clue #14 H has 7 neutrons

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Table of Contents Physical Science If8767 Answer Key Page 61

1. Understanding the eBook Physical Science If8767 Answer Key Page 61
 - The Rise of Digital Reading Physical Science If8767 Answer Key Page 61
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Answer Key Page 61
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Answer Key Page 61
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Answer Key Page 61
 - Personalized Recommendations
 - Physical Science If8767 Answer Key Page 61 User Reviews and Ratings
 - Physical Science If8767 Answer Key Page 61 and Bestseller Lists
5. Accessing Physical Science If8767 Answer Key Page 61 Free and Paid eBooks
 - Physical Science If8767 Answer Key Page 61 Public Domain eBooks
 - Physical Science If8767 Answer Key Page 61 eBook Subscription Services
 - Physical Science If8767 Answer Key Page 61 Budget-Friendly Options
6. Navigating Physical Science If8767 Answer Key Page 61 eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science If8767 Answer Key Page 61 Compatibility with Devices
 - Physical Science If8767 Answer Key Page 61 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Answer Key Page 61
 - Highlighting and Note-Taking Physical Science If8767 Answer Key Page 61
 - Interactive Elements Physical Science If8767 Answer Key Page 61
 8. Staying Engaged with Physical Science If8767 Answer Key Page 61
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Answer Key Page 61
 9. Balancing eBooks and Physical Books Physical Science If8767 Answer Key Page 61
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Answer Key Page 61
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science If8767 Answer Key Page 61
 - Setting Reading Goals Physical Science If8767 Answer Key Page 61
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science If8767 Answer Key Page 61
 - Fact-Checking eBook Content of Physical Science If8767 Answer Key Page 61
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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