

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is a redox reaction.



4.8 Write the half-cell reaction for the anode in the above cell.

4.9 Write the overall cell reaction for the above cell.

5.1 Write down the name of the reaction in the above diagram.

5.2 Write down the name of the reaction in the above diagram.



5.3 Write down the name of the reaction in the above diagram.

5.4 Write down the name of the reaction in the above diagram.

5.5 Write down the name of the reaction in the above diagram.

5.6 Write down the name of the reaction in the above diagram.

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Siddappa N.Byrareddy



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Table of Contents Physical Science Grd2013 Paper2 November Memo

1. Understanding the eBook Physical Science Grd2013 Paper2 November Memo
 - The Rise of Digital Reading Physical Science Grd2013 Paper2 November Memo
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grd2013 Paper2 November Memo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grd2013 Paper2 November Memo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grd2013 Paper2 November Memo
 - Personalized Recommendations
 - Physical Science Grd2013 Paper2 November Memo User Reviews and Ratings
 - Physical Science Grd2013 Paper2 November Memo and Bestseller Lists
5. Accessing Physical Science Grd2013 Paper2 November Memo Free and Paid eBooks

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 - Physical Science Grd2013 Paper2 November Memo eBook Subscription Services
 - Physical Science Grd2013 Paper2 November Memo Budget-Friendly Options
6. Navigating Physical Science Grd2013 Paper2 November Memo eBook Formats
- ePub, PDF, MOBI, and More
 - Physical Science Grd2013 Paper2 November Memo Compatibility with Devices
 - Physical Science Grd2013 Paper2 November Memo Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Grd2013 Paper2 November Memo
 - Highlighting and Note-Taking Physical Science Grd2013 Paper2 November Memo
 - Interactive Elements Physical Science Grd2013 Paper2 November Memo
8. Staying Engaged with Physical Science Grd2013 Paper2 November Memo
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grd2013 Paper2 November Memo
9. Balancing eBooks and Physical Books Physical Science Grd2013 Paper2 November Memo
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grd2013 Paper2 November Memo
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grd2013 Paper2 November Memo
- Setting Reading Goals Physical Science Grd2013 Paper2 November Memo
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grd2013 Paper2 November Memo
- Fact-Checking eBook Content of Physical Science Grd2013 Paper2 November Memo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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