Molecular Geometry Chart

# of Electron Groups	Number of Lone Pairs	Electron Pair Arrangement	Molecular Geometry	Approximate Bond Angles
2	0	linear	180"	180°
3	o	trigonal planar	1200	120°
	1	bent		<120°
4	o	tetrahedral	3	109.5°
	1	trigonal pyramid	3000	<109.5° (~107°)
	2	bent	3-100.5	<109.5°(~105°)
5	o	trigonal bipyramidal	13-38-5	90°, 120°
	1	see-saw		<90°, <120°
	2	T-structure	3	<90°
	3	linear	() I () () () () ()	180°
6	o	octahehral	833.00	90°, 90°
	1	square pyramidal	90° 33 90°	90°, <90°
	2	square planar	8:3:9:00	90°

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Linus Pauling, Roger Hayward

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