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Chapter 7 Chemical Reactions Section 7.1 Describing Reactions

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Name _____ Class _____ Date _____ Chapter 7 Chemical Reactions Section 7.1
Describe Changes in Reactions

Chemical Reactions - Purdue University College of AC

Chemical reactions produce substances with properties that are different from the reactants.

Chemical reactions are described by chemical equations. Example: The reaction between hydrogen and oxygen to form water is represented by the

equation: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

class 7 - Science - Ch9 - Physical and Chemical Changes



Chapter 9: Physical and Chemical Changes

9.1 Matter / Physical Changes / Chemical Changes / 9.2 Mixtures / 9.3 Acids and Bases

Understanding the physical and chemical changes that occur in matter.

How is boiling a solid a chemical reaction - Answers - 202

Answers.com - Questions - Science - Chemistry

Boiling is a physical reaction, not a chemical reaction. First off, there is no energy applied to change a substance's state (ice to water or liquid to gas) (phase change).

What happens to atoms in a chemical reaction?

Answers.com - Questions - Science - Chemistry

Chemical reactions involve atoms and molecules. Atoms combine together and split apart. For example, if carbon (C) is burnt in oxygen (O₂) to form carbon dioxide...

Section 7 Wordwise Answers

Albert A Gayle



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