SOLUTIONS TO PROBLEMS

Section 1.1 Standards of Length, Mass, and Time

No problems in this section

Section 1.2 Matter and Model-Building

From the figure, we may see that the spacing between diagonal planes is half the distance between diagonally adjacent atoms on a flat plane. This diagonal distance may be obtained from the Pythagorean theorem, $L_{\rm diag} = \sqrt{L^2 + L^2}$. Thus, since the atoms are separated by a distance L = 0.200 nm, the diagonal planes are separated by $\frac{1}{2}\sqrt{L^2 + L^2} = \boxed{0.141 \text{ nm}}$.

Section 1.3 Density and Atomic Mass

- *P1.2 Modeling the Earth as a sphere, we find its volume as $\frac{4}{3}\pi r^3 = \frac{4}{3}\pi \left(6.37\times10^6\ \mathrm{m}\right)^3 = 1.08\times10^{21}\ \mathrm{m}^3$. Its density is then $\rho = \frac{m}{V} = \frac{5.98\times10^{24}\ \mathrm{kg}}{1.08\times10^{21}\ \mathrm{m}^3} = \left[5.52\times10^3\ \mathrm{kg/m}^3\right]$. This value is intermediate between the tabulated densities of aluminum and iron. Typical rocks have densities around 2 000 to 3 000 kg/m³. The average density of the Earth is significantly higher, so higher-density material must be down below the surface.
- P1.3 With $V = (base area)(height) V = (\pi r^2)k$ and $\rho = \frac{m}{V}$, we have

$$\rho = \frac{m}{\pi r^2 h} = \frac{1 \text{ kg}}{\pi (19.5 \text{ mm})^2 (39.0 \text{ mm})} \left(\frac{10^9 \text{ mm}^3}{1 \text{ m}^3} \right)$$
$$\rho = \left[2.15 \times 10^4 \text{ kg/m}^3 \right].$$

- *P1.4 Let V represent the volume of the model, the same in $\rho = \frac{m}{V}$ for both. Then $\rho_{\text{inon}} = 9.35 \text{ kg/V}$ and $\rho_{\text{gold}} = \frac{m_{\text{gold}}}{V}$. Next, $\frac{\rho_{\text{gold}}}{\rho_{\text{inon}}} = \frac{m_{\text{gold}}}{9.35 \text{ kg}}$ and $m_{\text{gold}} = 9.35 \text{ kg} \left(\frac{19.3 \times 10^3 \text{ kg/m}^3}{7.86 \times 10^3 \text{ kg/m}^3} \right) = \boxed{23.0 \text{ kg}}$.
- PLS $V = V_a V_i = \frac{4}{3}\pi (r_2^3 r_1^3)$ $\rho = \frac{m}{V}$, so $m = \rho V = \rho (\frac{4}{3}\pi)(r_2^3 - r_1^3) = \frac{4\pi \rho (r_2^3 - r_1^3)}{3}$.

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L Darling-Hammond

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