

PHYSICAL SCIENCES (CHEMISTRY) GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2

TIME: 1 HOUR

(1)

(6)

(1)

(3)

(2)

(4)

VERIFICATION OF BOYLE'S LAW

INDIVIDUAL WORK (should be done under strict supervision).

- Write a scientific report which must include the following:
- 1.1 Aim
- 1.2 Apparatus and method (5)
- Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate and pV for each set of results and record your answers in the table.

Volume (V) cm³	Pressure (p) kPa	pV (°)
	+	
	+	

- 3. Analysis of the results:
- 3.1 Identify the following variables for the experiment:
 - Independent
 - Dependent
- Controlled

 3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus.

 (18)
- 4. Conclusion

55

- 4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.
- 4.2 Write down a conclusion for this experiment.

Questions and discussion:

- 5.1 Explain why you have to wait for a few minutes before you can take readings.
- 5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014

Thirumalaisamy P. Velavancorresponding

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Table of Contents Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014

- 1. Understanding the eBook Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - The Rise of Digital Reading Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Exploring Different Genres
 - $\circ\,$ Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Personalized Recommendations
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 User Reviews and Ratings
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 and Bestseller Lists
- 5. Accessing Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 Free and Paid eBooks
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 Public Domain eBooks
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 eBook Subscription Services
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 Budget-Friendly Options
- 6. Navigating Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 Compatibility with Devices
 - Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Highlighting and Note-Taking Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Interactive Elements Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
- 8. Staying Engaged with Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
- 9. Balancing eBooks and Physical Books Sekhukhune District Physical Science Common Test No 1 Grade12 Question

Paper 2014

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Sekhukhune District Physical Science Common Test No 1 Grade12
 Question Paper 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Setting Reading Goals Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - Fact-Checking eBook Content of Sekhukhune District Physical Science Common Test No 1 Grade12 Question Paper 2014
 - o Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - o Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - $\circ \ \ Integration \ of \ Multimedia \ Elements$
 - Interactive and Gamified eBooks

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