Section 6.1 Ionic Bonding

(pages 158-164)

This section describes the formation of looks hands and the properties of looks compounds.

Reading Strategy (2009) 158)

Sequencing. As you read, complete the concept map to show what happens to atoms during tonic bonding. For more information on this Bending Strategy, see the Beading and Study Skills in the Skills and Beforence Handbook at the end of your textbook.



Stable Electron Configurations (use 158)

- Describe the type of electron configuration that makes an atom stable and not likely to react. When in higher constant many lives of an electronic wife,
- 2. Describe an electron dot diagram. In matters are diagram to a material material and a material materials.

fonic Bonds (pages 189-191)

- By losing one valence electron, a sodium atom achieves the same electron arrangement as an atom of
- Circle the letter that states the result of a sodium atom transferring an electron to a chilorine atom.
 - a. Each atom ends up with a more stable electron orningement.
 - The so-dison atom becomes more stable, but the chilorine atom becomes less stable.
 - The chloring atom becomes more stable, but the sodium atom becomes less stable.
 - d. Each atom ends up with a less stable electron arrangement.
- Is the following sentence true or Table? An ion is an atom that has a net positive or negative electric charge.
- 7. An ion with a negative charge is called atm

Section 6 1 Ionic Bonding Worksheet Answer Key

Léo Pomar

Section 6 1 Ionic Bonding Worksheet	Answer Key:

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