

# Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. 

©

 What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example:  $C_2H_2$ 

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: H-H

An electron dot formula uses dots instead of lines to show shared pairs of electrons. Example: H:H

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl <sub>2</sub>		Note: The electron dot structures have dots in between each atom instead of lines.	:çı—çı:
b) H <sub>2</sub> S			:\$-н Н
c) Brl			ij <u> </u>

# **Section Review 7 Introduction To Bonding Answer Key**

Jianjun Gao

**Section Review 7 Introduction To Bonding Answer Key:** 

Whispering the Secrets of Language: An Psychological Quest through **Section Review 7 Introduction To Bonding Answer Key** 

In a digitally-driven world where monitors reign supreme and instant connection drowns out the subtleties of language, the profound secrets and mental nuances hidden within words usually go unheard. However, nestled within the pages of **Section Review 7 Introduction To Bonding Answer Key** a charming literary value sporting with natural thoughts, lies an exceptional quest waiting to be undertaken. Composed by an experienced wordsmith, this enchanting opus attracts viewers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within the very cloth of each word. Within the psychological depths with this poignant review, we can embark upon a sincere exploration of the book is core themes, dissect their charming writing fashion, and succumb to the strong resonance it evokes deep within the recesses of readers hearts.

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