

First/Second Semester B.E. Degree Examination, June/July 2013
Engineering Physics

Time: 3 hrs.

Max. Marks:100

Note: 1. Answer any FIVE full questions, choosing at least two from each part.

2. Answer all objective type questions only on OMR sheet page 5 of the answer booklet.

3. Answer to objective type questions on sheets other than OMR will not be valued.

4. Physical Constants : Planck's constant, $h = 6.63 \times 10^{-34}$ Js, Electron charge, $e = 1.6 \times 10^{-19}$ C
Electron mass, $m = 9.11 \times 10^{-31}$ kg, Velocity of light, $C = 3 \times 10^8$ mS⁻¹**PART - A**

- 1 a. Choose the correct answers for the following : (04 Marks)
- If red and blue stars emits radiations of continuous wavelengths, then according to Wien's displacement law.
 A) Blue star is hotter than red star
 B) Red star is hotter than blue star
 C) Both stars are at same temperature
 D) Difficult to conclude.
 - The expression for de-Broglie wavelength for an electron under an accelerating potential V is,
 A) $\frac{12.26}{\sqrt{V}}$ m
 B) $\frac{12.26}{\sqrt{V}}$ Å
 C) $\frac{12.26}{\sqrt{V}}$ nm
 D) $\frac{12.26}{\sqrt{V}}$ µm
 - A particle moves with velocity 3×10^8 ms⁻¹. The wavelength associated with it is 1 nm. Then group velocity of the particle is,
 A) 3×10^8 mS⁻¹
 B) 3×10^{10} mS⁻¹
 C) 3×10^6 mS⁻¹
 D) 1.5×10^6 mS⁻¹
 - According to the Compton effect, the wavelength of X-rays scattered at an angle greater than zero,
 A) Decreases
 B) Doesn't change
 C) Increases
 D) None of these
- b. Derive an expression for group velocity on the basis of superposition of waves. Also obtain the relation between group velocity and phase velocity. (08 Marks)
- c. Show that Planck's law reduces to Wien's law and Rayleigh-Jeans law under certain conditions. (05 Marks)
- d. Calculate the de-Broglie wavelength associated with an electron of energy 1.5 eV. (03 Marks)
- 2 a. Choose the correct answers for the following : (04 Marks)
- The energy of the lowest state in one dimensional potential box of length $a = 1$ unit is,
 A) $\frac{h^2}{8m}$
 B) zero
 C) $\frac{h^2}{4ma^2}$
 D) $\frac{h^2}{2ma^2}$
 - For a particle which is not bound to any system and is free, the energy eigen value is,
 A) zero
 B) finite but not quantized
 C) infinity
 D) finite but quantized
 - If the uncertainty in the position of a particle is equal to its de-Broglie wavelength then uncertainty in its momentum will be,
 A) $\Delta P \geq \frac{h}{4\pi}$
 B) $\Delta P \geq \frac{h}{2\pi}$
 C) $\Delta P \geq \frac{P}{4\pi}$
 D) $\Delta P \geq \frac{h}{P}$
 - For an electron to be present inside the nucleus of an atom the uncertainty in the position of the electron must be,
 A) more than or equal to the radius of the nucleus
 B) more than or equal to the diameter of the nucleus.
 C) more than the diameter of the nucleus
 D) less than or equal to the diameter of the nucleus.



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Table of Contents Physics 6phmarkscheme June 2013

1. Understanding the eBook Physics 6phmarkscheme June 2013
 - The Rise of Digital Reading Physics 6phmarkscheme June 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Physics 6phmarkscheme June 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics 6phmarkscheme June 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physics 6phmarkscheme June 2013
 - Personalized Recommendations
 - Physics 6phmarkscheme June 2013 User Reviews and Ratings
 - Physics 6phmarkscheme June 2013 and Bestseller Lists
5. Accessing Physics 6phmarkscheme June 2013 Free and Paid eBooks

- Physics 6phmarkscheme June 2013 Public Domain eBooks
- Physics 6phmarkscheme June 2013 eBook Subscription Services
- Physics 6phmarkscheme June 2013 Budget-Friendly Options
- 6. Navigating Physics 6phmarkscheme June 2013 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physics 6phmarkscheme June 2013 Compatibility with Devices
 - Physics 6phmarkscheme June 2013 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics 6phmarkscheme June 2013
 - Highlighting and Note-Taking Physics 6phmarkscheme June 2013
 - Interactive Elements Physics 6phmarkscheme June 2013
- 8. Staying Engaged with Physics 6phmarkscheme June 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics 6phmarkscheme June 2013
- 9. Balancing eBooks and Physical Books Physics 6phmarkscheme June 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics 6phmarkscheme June 2013
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physics 6phmarkscheme June 2013
 - Setting Reading Goals Physics 6phmarkscheme June 2013
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physics 6phmarkscheme June 2013
 - Fact-Checking eBook Content of Physics 6phmarkscheme June 2013
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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