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**Emilie Sanchez**



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web jul 12 2019 when we neutralize a weak acid with a strong base we get a salt that contains the conjugate base of the weak acid this conjugate base is usually a weak base for example sodium acetate  $\text{NaCH}_3\text{CO}_2$  is a salt formed by the reaction of the weak acid acetic acid with the strong base sodium hydroxide

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web summary an arrhenius acid produces  $\text{H}^+$  and an arrhenius base produces  $\text{OH}^-$  in aqueous solutions acids taste sour may sting and neutralize bases bases taste bitter feel slippery and neutralize acids naming acids binary acids contain a single anion  $\text{HX}$  to name hydro anion with  $\text{H}^+$  end acid

**acids and bases practice problems chemistry steps** - Jul 14 2023

web definition of acids and bases identifying them along with the conjugate acid and conjugate base ph of strong and weak acids and bases the ph and poh relationship  $K_a$  and  $K_b$  p  $K_a$  and  $K_a$  correlation the acid base properties of salts identifying the acidity of the salt based on the acid and the base it is prepared from and calculating the