

Chapter 7: Chemical Reaction Guided Notes

I. Chemical Changes in Matter

A. Chemical Reaction

1. A change in which one or more substances are converted to _____ substances.

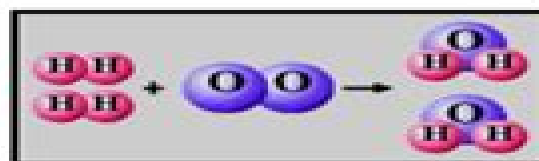


B. Law of Conservation of Matter

1. In a chemical reaction, matter is not _____ or _____.
2. Atoms can only be _____.
3. Discovered by Lavoisier.

_____ Hydrogen

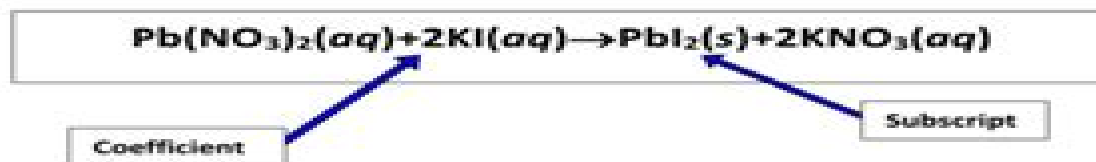
_____ Oxygen



_____ Hydrogen

_____ Oxygen

C. Chemical Equations



1. _____ - # of units of each substance.
2. Individual atom = **atom**
 $2\text{Mg} \Rightarrow 2$ _____ of magnesium
3. Covalent substance = **molecule**
 $3\text{CO}_2 \Rightarrow 3$ _____ of carbon dioxide
4. Ionic substance = **unit**
 $4\text{MgO} \Rightarrow 4$ _____ of magnesium oxide

SYMBOL	MEANING
\longrightarrow	produces, forms
+	plus, and
(s)	solid
(l)	liquid
(g)	gas
(aq)	aqueous (solid dissolved in water)
Δ	the reactants are heated

Notetaking Guide Answer Keychemical Reactions

Lawrie Ryan,Keith Johnson



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web if the rate of the forward reaction decreases increases stays the same more products are formed in a chemical reaction at equilibrium the rate of the forward reaction is equal to the rate of the reverse reaction if the rate of the forward reaction increases more products are formed

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